Collaborative Testing Services, Inc FORENSIC TESTING PROGRAM

# Serial Number Restoration Test No. 16-5250 Summary Report 

This test was sent to 344 participants. Each participant received a sample pack containing a piece of zinc bar stock which had been stamped with a six character serial number which was then obliterated. Also included was a piece of aluminum bar stock intended as a standard for the size, shape and positioning of the stamped characters. Participants were asked to restore the obliterated serial number. Data were returned from 302 participants ( $88 \%$ response rate) and are compiled into the following tables:

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## Manufacturer's Information

Each sample set consisted of a piece of zinc bar stock that contained an obliterated serial number (Item 1) and a piece of aluminum bar stock intended as a standard for the size, shape, and positioning of the stamped digits. Participants were requested to attempt to restore the obliterated serial number utilizing their laboratory restoration methodologies and report the recovered serial number. The serial number to be restored consisted of 6 characters (D7J274).

SAMPLE PREPARATION-
Each sample set contained a piece of $1^{\prime \prime} \times 1^{1 / 4} \times 2.5^{\prime \prime}$ zinc bar stock that was stamped using a punch press. The stamp consisted of 6 characters (D7J274) that are $1 / 8^{\prime \prime}$ in height. The serial number was then obliterated by removing material from the bar stock using a vertical milling machine. A consistent amount of material was removed from each piece of bar stock.

A piece of aluminum bar stock was also included in the sample set and was intended as a standard for size, shape, and positioning of the stamped alphanumeric characters used in the serial number. The alphanumeric characters are digits $0-9$ and letters $A-F, H, J, K$ and $N$ that are $1 / 8^{\prime \prime}$ in height.

SAMPLE SET ASSEMBLY: An Item 1 was enclosed in chip board and the sides taped to securely contain the sample. The aluminum standard was wrapped in chip board and both were placed in their respective pre-labeled envelopes. Each sample pack was packaged with an Item 1 and aluminum standard. This process was repeated until all of the sample packs were prepared. Once verification was completed, all sample packs were sealed with a piece of evidence tape and initialed "CTS."

VERIFICATION: Three predistribution laboratories restored the obliterated six character serial number and reported "D7J274". All three laboratories used a chemical restoration method for recovery.

## Summary Comments

This test was designed to allow participants to assess their proficiency in the restoration of an obliterated serial number. Participants were provided with a piece of zinc bar stock that contained an obliterated serial number (ltem 1) and a piece of aluminum bar stock intended as a standard for the size, shape and positioning of the stamped characters. Participants were requested to restore the obliterated serial number utilizing their laboratory recovery methodologies and report the recovered serial number. The serial number to be restored consisted of 6 characters (D7J274). (Refer to Manufacturer's Information for preparation details.)

Of the 302 responding participants in Table 1: "Recovered Characters", 263 (87\%) recovered the six characters consistent with the Manufacturer's Information. Twelve participants made a partial recovery of the expected characters, and eleven participants made a full or partial recovery with one or two reported characters that were different from, but similar in shape to, the expected characters. The remaining sixteen participants did not recover any of the characters.

In Table 3: "Preparation Methods", the majority of participants used polishing or sanding to prepare their sample. In Table 4: "Recovery Methods", a vast majority of participants used chemical processing for the serial number restoration.

## Recovered Characters

Please record the restored characters below.
TABLE 1

| WebCode | Character 1 | Character 2 | Character 3 | Character 4 | Character 5 | Character 6 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 26JQKN | D | 7 | J | 2 | 7 | 4 |
| 27EALP | D | 7 | J | 2 | 7 | 4 |
| 2ACVB8 | D | 7 | J | 2 | 7 | 4 |
| 2B8DD9 | D | 7 | J | 2 | 7 | 4 |
| 2BNLMW | D | 7 | J | 2 | 7 | 4 |
| $2 C J 4 N X$ | ? | ? | ? | ? | ? | ? |
| 2NTQPH | D | 7 | J | 2 | 7 | 4 |
| 2P7FRV | D | 7 | J | 2 | 7 | 4 |
| 2QJRWR | D | 7 | J | 2 | 7 | 4 |
| 2QXCGP | D | 7 | J | 2 | 7 | 4 |
| 2QZ9ME | D | 7 | J | 2 | 7 | 4 |
| 2WKBKH | D | 7 | J | 2 | 7 | 4 |
| 2XW2XT | D | 7 | J | 2 | 7 | 4 |
| 2ZQHH3 | D | 7 | J | 2 | 7 | 4 |
| $32 \mathrm{BX2A}$ | 0 | 7 | J | 2 | 7 | 4 |
| 33R6BX | D | 7 | J | 2 | 7 | 4 |
| 34923X | D | 7 | J | 2 | 7 | 4 |
| 3836JH | D | 7 | J | 2 | 7 | 4 |
| $3 \mathrm{B64WW}$ | D | 7 | J | 2 | 7 | 4 |
| $3 B Z Q B L$ | D | 7 | J | 2 | 7 | 4 |
| 3D6C27 | D | 7 | J | 2 | 7 | 4 |
| 3MUZXU | D | 7 | J | 2 | 7 | 4 |
| $3 N P H Z V$ | D | 7 | J | 2 | 7 | 4 |

TABLE 1

| WebCode | Character 1 | Character 2 | Character 3 | Character 4 | Character 5 | Character 6 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 3PK7EK | D | 7 | J | 2 | 7 | 4 |
| 3QFPFL | D | 7 | J | 2 | 7 | 4 |
| 3RTGBN | D | 7 | J | 2 | 7 | 4 |
| 3UD7U2 | D | 7 | J | 2 | 7 | 4 |
| 3ZJYPW | D | 7 | J | 2 | 7 | 4 |
| 438NW | D | 7 | J | 2 | 7 | 4 |
| 477ZQV | D | 7 | J | 2 | 7 | 4 |
| 482HTW | D | 7 | J | 2 | 7 | 4 |
| 4GK94M | D | 7 | J | 2 | 7 | 4 |
| 4LE34A | D | 7 | J | 2 | 7 | 4 |
| 4R2QYE | D | 7 | J | 2 | 7 | 4 |
| 4TFF2R | D | 7 | J | 2 | 7 | 4 |
| 4TG2TD | D | 7 | J | 2 | 7 | 4 |
| 4UVG6U | D | 7 | J | 2 | 7 | 4 |
| 4WDWEL | / | 1 | 1 | 1 | 1 | 1 |
| 4XR3XH | D | 7 | J | 2 | 7 | 4 |
| 672RCF | D | 7 | J | 2 | 7 | 4 |
| 6DFXMD | D | 7 | J | 2 | 7 | 4 |
| 6DYT6E | D | 7 | J | 2 | 7 | 4 |
| 6EYJLV | D | 7 | J | 2 | 7 | 4 |
| 6HCJJM | D | 7 | J | 2 | 7 | 4 |
| 6KKLGE | D | 7 | J | 2 | 7 | 4 |
| 6N3UTK | D | 7 | J | 2 | 7 | 4 |
| 6RNJDX | D | 7 | J | 2 | 7 | 4 |
| 6TVVUY | D | 7 | J | 2 | 7 | 4 |

TABLE 1

## WebCode

Character 1 Character 2 Character 3 Character 4 Character 5 Character 6

| 6XBJEW | D | 7 | J | 2 | 7 | 4 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| 78VZW6 | D | 7 | J | 2 | 7 | 4 |
| 7WCH7J | D | 7 | J | 2 | 7 | 4 |

7WPCHD

| 7YFDPM | D | 7 | J | 2 | 7 | 4 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 86Y4M2 | D | 7 | J | 2 | 7 | 4 |
| 88QYAA | D | 7 | J | 2 | 7 | 4 |
| 8ATXMN | D | 7 | J | 2 | 7 | 4 |
| 8DEM6F | D | 7 | 3 | 2 | 7 | 4 |
| 8GC4GA | D | 7 | J | 2 | 7 | 4 |
| 8XYGXM | D | 7 | * | 2 | 7 | 4 |
| 92WHBP | D | 7 | 0 | 2 | 7 | 4 |
| 94H7TG | D | 7 | J | 2 | 7 | 4 |
| 96T7RW | D | 7 | J | 2 | 7 | 4 |
| 97A3JV | D | 7 | J | 2 | 7 | 4 |
| 97NPTX | D | 7 | J | 2 | 7 | 4 |
| 99XRPU | D | 7 | J | 2 | 7 | 4 |
| 9E3B6M | D | 7 | J | 2 | 7 | 4 |
| 9EKJTG | D | 7 | J | 2 | 7 | 4 |
| 9FY4LR | D | 7 | J | 2 | 7 | 4 |
| 9K4Y98 | D | 7 | J | 2 | 7 | 4 |
| 9KN9QN | D | 7 | J | 2 | 7 | 4 |
| 9L3XU2 | D | 7 | * | * | 7 | 4 |
| 9MVQPK | D | 7 | J | 2 | 7 | 4 |
| 9N8HJM | D | 7 | J | 2 | 7 | 4 |

TABLE 1

## WebCode

## Character 1 Character 2 Character 3 Character 4 Character 5 Character 6

| 9 Q3Y6W | D | 7 | J | 2 | 7 | 4 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| 9 QBD4P | D | 7 | J | 2 | 7 | 4 |

9T3CF3

| 9XYRDQ | D | 7 | J | 2 | 7 | 4 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| 9Y76NC | D | 7 | J | 2 | 7 | 4 |
| A73MD9 | D | 7 | J | 2 | 7 | 4 |
| A8ZNN7 |  |  |  |  |  |  |


| AA4JAV | D | 7 | J | 2 | 7 | 4 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| ACA28Q | D | 7 | J | 2 | 7 | 4 |
| ADMNRQ | D | 7 | J | 2 | 7 | 4 |
| AEYADL | D | 7 | J | 2 | 7 | 4 |
| AJDY2N | D | 7 | J | 2 | 7 | 4 |
| AJM3G9 | D | 7 | J | 2 | 7 | 4 |
| AJRJE7 | D | 7 | J | 2 | 7 | 4 |
| ALYRUH | D | 7 | J | 2 | 7 | 4 |

AVGH79

| AXBZT4 | D | 7 | J | 2 | 7 | 4 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| B284XV | D | 7 | J | 2 | 7 | 4 |
| B4ELVR | D | 7 | J | 2 | 7 | 4 |
| B67GKL | D | 7 | J | 2 | 7 | 4 |
| BAHHPP | D | 7 | J | 2 | 7 | 4 |
| BHFXRQ | D | 7 | J | 2 | 7 | 4 |
| BWCAZU | D | 7 | J | 2 | 7 | 4 |
| BWXQFX | D | 7 | J | 2 | 7 | 4 |
| BZCKVU | D | 7 | J | 2 | 7 | 4 |

WebCode

## Character 1 Character 2 Character 3 Character 4 Character 5 Character 6

| C4T88E | D | 7 | J | 2 | 7 | 4 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C4VPEP | D | 7 | J | 2 | 7 | 4 |
| C6Q9FR | D | 7 | J | 2 | 7 | 4 |
| CAGUZY | D | 7 | J | 2 | 7 | 4 |
| CAV438 | D | 7 | J | 2 | 7 | 4 |
| CCKPJ2 | D | 7 | J | 2 | 7 | 4 |
| CGG7YG | D | 7 | J | 2 | 7 | 4 |
| CNGTNV | D | 7 | J | 2 | 7 | 4 |
| CYJQW6 | D | 7 | J | 2 | 7 | 4 |
| CZPHF6 | D | 7 | J | 2 | 7 | 4 |
| D8MK7N | D | 7 | J | 2 | 7 | 4 |
| DDABPX | D | 7 | J | 2 | 7 | 4 |
| DFFTQE | D | 7 | J | 2 | 7 | 4 |
| DGXMRN | D | 7 | J | 2 | 7 | 4 |
| DHNT8E | D | 7 | J | 2 | 7 | 4 |

DRF6JV

| DTLUKJ | D | 7 | J | 2 | 7 | 4 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| DVV2MC | D | 7 | J | 2 | 7 | 4 |
| E2WBVP | D | 7 | J | 2 | 7 | 4 |
| E6Y96W | D | 7 | J | 2 | 7 | 4 |
| E77NDN | D | 7 | J | 2 | 7 | 4 |
| E7QG4Z | D | 7 | J | 2 | 7 | 4 |
| E9WUUE | D | 7 | J | 2 | 7 | 4 |
| EHZ63P | D | 7 | J | 2 | 7 | 4 |
| EJTXX9 | D | 7 | J | 2 | 7 | 4 |

TABLE 1

## WebCode

## Character 1 Character 2 Character 3 Character 4 Character 5 Character 6

| EJYBJJ | D | 7 | J | 2 | 7 | 4 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| EP43WM | D | 7 | J | 2 | 7 | 4 |
| EQU347 | D | 7 | J | 2 | 7 | 4 |

ERB8L6

| ERPVPH | D | 7 | J | 2 | 7 | 4 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| EUFYDH | D | 7 | J | 2 | 7 | 4 |
| F33DUT | D | 7 | J | 2 | 7 | 4 |
| F4JCCX | D | 7 | J | 2 | 7 | 4 |
| F8VGLC | D | 7 | J | 2 | 7 | 4 |
| F9QZMD | D | 7 | J | 2 | 7 | 4 |
| F9TPUW | D | 7 | J | 2 | 7 | 4 |
| FM9WK2 | D | 7 | J | 2 | 7 | 4 |
| FR6CHP | D | 7 | J | 2 | 7 | 4 |
| FTX96X | D | 7 | J | 2 | 7 | 4 |
| FV4TTX | D | 7 | J | 2 | 7 | 4 |
| FYLT9D | D | 7 | 3 | 2 | 7 | 4 |
| G7GCPZ | D | 7 | J | 2 | 7 | 4 |
| G88KP3 | D | 7 | J | 2 | 7 | 4 |
| GA4PQQ | * | 7 | * | 2 | 7 | 4 |
| GBXHLA | D | 7 | J | 2 | 7 | 4 |
| GCAAGC | D | 7 | J | 2 | 7 | 4 |
| GF7XVX | D | 7 | J | 2 | 7 | 4 |
| GG8LKN | D | 7 | J | 2 | 7 | 4 |

GJAW7Q
GLDFEK D

## WebCode

## Character 1 Character 2 Character 3 Character 4 Character 5 Character 6

| GUMWZY | D | 7 | ? | 2 | 7 | 4 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| GVZPED | D | 7 | J | 2 | 7 | 4 |
| GWW7QB | D | 7 | J | 2 | 7 | 4 |
| GWYJMJ | D | 7 | J | 2 | 7 | 4 |
| H24RGT | D | 7 | 0 | 2 | 7 | 4 |
| H7CC36 | D | 7 | J | 2 | 7 | 4 |
| H8BHJY | D | 7 | J | 2 | 7 | 4 |
| HC7ATG | D | 7 | J | 2 | 7 | 4 |
| HX6B7K | D | 7 | J | 2 | 7 | 4 |
| J46GW8 | D | 7 | J | 2 | 7 | 4 |
| J4M4UN | D | 7 | J | 2 | 7 | 4 |
| J6ZZX9 | D | 7 | J | 2 | 7 | 4 |
| JTFAYL | D | 7 | J | 2 | 7 | 4 |
| JEL2F4 | D | 7 | J | 2 | 7 | 4 |
| JEQCKM | D | 7 | ? | ? | 7 | 4 |
| JGQNVT | D | 7 | ? | 2 | 7 | 4 |
| JKEZWJ | D | 7 | J | 2 | 7 | 4 |
| JKQZQT | D | 7 | J | 2 | 7 | 4 |
| JLLHTU | D | 7 | J | 2 | 7 | 4 |
| JNECRJ | D | 7 | J | 2 | 7 | 4 |
| JP7B4W | D | 7 | J | 2 | 7 | 4 |
| JRXGBF | D | 7 | 3 | 2 | 7 | 4 |
| JT99GB | D | 7 | J | 2 | 7 | 4 |
| JYWTKQ | D | 7 | J | 2 | 7 | 4 |

TABLE 1
WebCode
Character 1 Character 2 Character 3 Character 4 Character 5 Character 6

| K3K6MZ | D | 7 | J | 2 | 7 | 4 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| K4KTEC | D | 7 | J | 2 | 7 | 4 |
| K4ZEND | D | 7 | J | 2 | 7 | 4 |
| K7PK44 | D | 7 | J | 2 | 7 | 4 |
| K9V7WB | D | 7 | J | 2 | 7 | 4 |
| KALAKB | D | 7 | J | 2 | 7 | 4 |
| KCV792 | D | 7 | J | 2 | 7 | 4 |
| KD8UR2 | D | 7 | J | 2 | 7 | 4 |
| KRT9VZ | D | 7 | J | 2 | 7 | 4 |
| KXFWQ7 | D | 7 | J | 2 | 7 | 4 |
| KZLGFQ |  |  |  |  |  |  |
| L36WLC | D | 7 | J | 2 | 7 | 4 |
| L4HKM4 | D | 7 | J | 2 | 7 | 4 |
| LG4ZQ4 | D | 7 | J | 2 | 7 | 4 |
| LGNLG9 | D | 7 | J | 2 | 7 | 4 |
| LGPBMQ | D | 7 | J | 2 | 7 | 4 |
| LHVXXA | D | 7 | J | 2 | 7 | 4 |
| LJQGZB | D | 7 | J | 2 | 7 | 4 |
| LN3JQD | D | 7 | J | 2 | 7 | 4 |
| LQD8M7 | D | 7 | J | 2 | 7 | 4 |
| LV4JLA | D | 7 | J | 2 | 7 | 4 |
| LV92K8 | D | 7 | J | 2 | 7 | 4 |
| LVJZG3 | D | 7 | J | 2 | 7 | 4 |
| LVPC3D | D | 7 | J | 2 | 7 | 4 |
| M4TPUZ | D | 7 | J | 2 | 7 | 4 |

## TABLE 1

## WebCode

## Character 1 Character 2 Character 3 Character 4 Character 5 Character 6

| M4Y2FC | D | 7 | J | 2 | 7 | 4 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| M6P8W3 | D | 7 | J | 2 | 7 | 4 |
| M73WYE | D | 7 | J | 2 | 7 | 4 |


| M8UVBR | D | 7 | J | 2 | 7 | 4 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| MD7UTC | D | 7 | J | 2 | 7 | 4 |


| MGHQB8 | D | 7 |
| :--- | :--- | :--- |

TABLE 1

## WebCode

## Character 1 Character 2 Character 3 Character 4 Character 5 Character 6

| PL4HDK | D | 7 | J | 2 | 7 | 4 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |

PP8ZPV

| PWLY6R | D | 7 | ? | 2 | 7 | 4 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| PZKB4D | D | 7 | J | 2 | 7 | 4 |
| Q2A4B2 | D | 7 | 0 | 2 | 7 | 4 |
| Q7TYET | D | 7 | J | 2 | 7 | 4 |
| Q9XLV4 | D | 7 | J | 2 | 7 | 4 |
| QA9NH3 | D | 7 | J | 2 | 7 | 4 |
| QAT6X7 | D | 7 | J | 2 | 7 | 4 |
| QD9ZD3 | D | 7 | J | 2 | 7 | 4 |
| QLYKGF | * | * | * | * | 7 | 4 |
| QRPVRE | D | 7 | J | 2 | 7 | 4 |
| QUE4YU | D | 7 | J | 2 | 7 | 4 |
| QY82Q7 | D | 7 | J | 2 | 7 | 4 |
| R6TFRH | D | 7 | J | 2 | 7 | 4 |
| RD7CU4 | D | 7 | J | 2 | 7 | 4 |
| RDLTPW | D | 7 | J | 2 | 7 | 4 |
| RKPVUW | D | 7 | J | 2 | 7 | 4 |
| RL43FE | D | 7 | J | 2 | 7 | 4 |
| RNQRLD | D | 7 | J | 2 | 7 | 4 |
| RPG3C7 | 0 | 7 | * | 2 | 7 | 4 |
| RR2QUX | D | 7 | 3 | 2 | 7 | 4 |
| RR79TW | D | 7 | J | 2 | 7 | 4 |
| RZWX6Q | D | 7 | J | 2 | 7 | 4 |
| T6PDDX | D | 7 | J | 2 | 7 | 4 |

## TABLE 1

## WebCode

## Character 1 Character 2 Character 3 Character 4 Character 5 Character 6

| TC8L4F | D | 7 | $?$ | 2 | 7 | 4 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |

TKZ48P

| TQ7MH2 | D | 7 | J | 2 | 7 | 4 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| TXN2G2 | ? | ? | ? | 2 | 7 | 4 |
| TXQQNJ | D | 7 | J | 2 | 7 | 4 |
| U2YFFQ | D | 7 | J | 2 | 7 | 4 |
| U8BEXC | D | 7 | J | 2 | 7 | 4 |
| UDWAYF | D | 7 | J | 2 | 7 | 4 |
| UJ34VW | D | 7 | J | 2 | 7 | 4 |


| UVWA2N | D | 7 | J | 2 | 7 | 4 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| UYWKWN | D | 7 | J | 2 | 7 | 4 |
| UZNR37 | D | 7 | J | 2 | 7 | 4 |
| V62RZX | D | 7 | J | 2 | 7 | 4 |
| VE976K | D | 7 | J | 2 | 7 | 4 |
| VJZQV8 | D | 7 | J | 2 | 7 | 4 |


| VM3CZ8 | D | 7 |  | 2 | 7 | 4 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| VP6ABZ | D | 7 | J | 2 | 7 | 4 |
| VRA33D | D | 7 | J | 2 | 7 | 4 |
| VYBGNE | D | 7 | J | 2 | 7 | 4 |
| VZ68UT | D | 7 | J | 2 | 7 | 4 |
| W2F8QM | D | 7 | J | 2 | 7 | 4 |
| W4LRF9 | D | 7 | J | 2 | 7 | 4 |
| W8L3AM | D | 7 | J | 2 | 7 | 4 |
| WBJJMG | D | 7 | J | 2 | 7 | 4 |

TABLE 1
WebCode
Character 1 Character 2 Character 3 Character 4 Character 5 Character 6

| WCARLK | D | 7 | J | 2 | 7 | 4 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| WD8VN9 |  |  |  |  |  |  |


| WFA3HK | D | 7 | J | 2 | 7 | 4 |
| :--- | :---: | :--- | :--- | :--- | :--- | :--- |
| WGAUYZ | D | 7 | J | 2 | 7 | 4 |
| WL22EN | D | 7 | J | 2 | 7 | 4 |
| WL79U3 | D | 7 | J | 2 | 7 | 4 |
| WQWTP7 | D | 7 | J | 2 | 7 | 4 |
| WRRBR8 | D | 7 | J | 2 | 7 | 4 |
| WTK8DF | D | 7 | J | 2 | 7 | 4 |
| WTMY7W | D | 7 | J | 2 | 7 | 4 |
| WXJHC2 | D | 7 | J | 2 | 7 | 4 |
| XA4J3R | D | 7 | J | 2 | 7 | 4 |
| XNM3Z3 | D | 7 | J | 2 | 7 | 4 |
| XQEYNW | D | 7 | J | 2 | 7 | 4 |

Y8DM7L

| Y96TC4 | D | 7 | J | 2 | 7 | 4 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| YBWP2X | D | 7 | J | 2 | 7 | 4 |
| YKDLTH | D | 7 | J | 2 | 7 | 4 |
| YKNYU2 | D | 7 | J | 2 | 7 | 4 |
| YMPC8T | D | 7 | J | 2 | 7 | 4 |
| YXBFER | D | 7 | J | 2 | 7 | 4 |
| Z9JKYT | D | 7 | J | 2 | 7 | 4 |
| ZAXZG2 | D | 7 | J | 2 | 7 | 4 |
| ZDHU9V | D | 7 | J | 2 | 7 | 4 |
| ZE9ZPL | D | 7 | J | 2 | 7 | 4 |

## TABLE 1

WebCode Character 1 Character 2 Character 3 Character 4 Character 5 Character 6

| ZJMXW9 | D | 7 | J | 2 | 7 | 4 |
| :--- | :--- | :--- | :--- | :--- | :--- | :--- |
| ZL47ZE |  |  |  |  |  |  |
| ZX3V4U | D | 7 | J | 2 | 7 | 4 |
| ZYVNYD | D | 7 | J | 2 | 7 | 4 |
| ZZWCN4 | D | 7 | J | 2 | 7 | 4 |

Response Summary Participants: 302

Character 1 Character 2 Character 3 Character 4 Character 5 Character 6

| Consensus | D | 7 | J | 2 | 7 | 4 |
| :--- | :---: | :---: | :---: | :---: | :---: | :---: |
| Number | 280 | 284 | 265 | 283 | 286 | 286 |
| Percent | $92.7 \%$ | $94.0 \%$ | $87.7 \%$ | $93.7 \%$ | $94.7 \%$ | $94.7 \%$ |

## Conclusions

## TABLE 2

## Conclusions

26JQKN The obliterated serial number on the bar (ltem 1) was mechanically and chemically processed and restored to read D7J274.

27EALP The obliterated serial number on zinc bar Item 1, was restored to read: D7J274
2ACVB8 Examination of Item 1 revealed an obliterated area on one side. Standard chemical restoration techniques revealed the following characters: D7J274.

2B8DD9 After application of the electro-acid etching process, I determined the serial number of the bar stock as D7J274.

2BNLMW Using a combination of mechanical and chemical restoration techniques, the serial number was fully restored to read: D7J274.

2CJ4NX Examination of the submitted bar stock revealed the serial number to have been obliterated. Physical and chemical restoration attempts were unsuccessful.

2NTQPH Attempts to restore the obliterated serial number on the piece of zinc bar stock, Item 1, were successful. The restored serial number is D7J274.

2P7FRV The obliterated number on Item 1 was examined microscopically, cleaned and chemically restored to read D7J274.

2QJRWR Serial number restoration techniques were applied to the submitted zinc bar (ltem 1). The serial number was determined to be D 7 J 274 .

2QXCGP The obliterated serial number on Item 1 was chemically processed and restored to read "D7J274".
2QZ9ME Examination and chemical processing of item 1 restored the original serial number, which was determined to be "D7J274".

2WKBKH The obliterated serial number was restored to read D7J274.
2XW2XT Using standard laboratory techniques, the serial number on Item \#1 was restored to read "D 7 J 27 4".

2ZQHH3 A series of characters were restored using a Forensic procedure. The restored characters were similar in height, font and stylisation when compared to the supplied "standard".

32BX2A The restoration of the area presenting alteration was done and it was possible to achieve the complete restoration where the alphanumeric sequence 07J274 was observed.

33R6BX Standard serial number restoration techniques revealed the following characters: D7J274.
34923X Physical and chemical processing restored the original obliterated serial number to read "D7J274".
3836JH SERIAL NUMBER FULLY RESTORED ON 3/25/16 TO READ D7J274
3B64WW Using chemical etchants the obliterated number on the Item 1 bar stock was restored to read D7J274.
3BZQBL Using chemical etching techniques, the serial number was restored to read D7J274.

## Conclusions

3D6C27 Examination of Item 1 revealed an obliterated area. Standard chemical restoration procedures were utilized and characters "D7J274" were restored.

3MUZXU Item 1 was microscopically examined. The obliterated number on the Item 1 piece of bar stock was polished and chemically restored to reveal the serial number D7J274.

3NPHZV Using standard laboratory serial number restoration techniques, an attempt was made to restore the serial number with the following results: Serial Number: D 7 J 274 was restored on Item 1.

3PK7EK THE SERIAL NUMBER PRINTED ON THE PRINCIPAL PROOF ITEM OR PHYSICAL EVIDENCE IN MATTER (ZINC BAR STOCK) IS D7J274 THAT WERE OBTAINED THROUGH THE RESTORING PROCESS OF THE SERIAL NUMBER; THE PRINCIPAL PROOF ÍTEM OR PHYSICAL EVIDENCE BELONGS TO THE TEST RESEARCH CTS N ${ }^{\circ} 16-5250$. SEE PHOTOGRAPHIC ANNEX.

3QFPFL The restoration techniques applied allowed the identification of the previously deleted serial number "D7J274"

3RTGBN The obliterated serial number was restored to read: "D7J274".
3UD7U2 Application of scientific techniques for the retrieval of the manufacturer's originally applied serial number revealed it to be D7J274.

3ZJYPW The serial number had been obliterated from the metal bar stock. Serial number restoration technique via acid etching yielded the serial number of D7J274.

438NW Item 1 was a 0.175 inch thick piece of zinc bar stock, measuring 2.300 by 0.954 inches, with a milled out area on one face. This area was cleaned, polished, and chemically etched in an attempt to restore the obliterated serial number. A serial number of D7J274 was successfully restored.

477ZQV The serial number was restored to read D7J274.
482HTW The serial number was restored to read D7J274.

4GK94M The serial number of the "zinc bar stock" was restored to read D 7 J 27 . The "Aluminum Standard" was not further examined.

4LE34A The serial number of the zinc bar stock, described in Item 1, was restored and corresponds to: D7J274.

4R2QYE Item 1 obliterated serial number located in the center of the sample was fully restored after examination and chemical processing. The serial number was determined to be "D7J274".

4TFF2R Serial Number restoration techniques were applied to Item 1. The Serial Number was determined to be D7J274.

4TG2TD Serial number restoration procedures were performed on the ltem 1 bar stock and the obliterated serial number was restored to read D7J274.

4UVG6U Using standard serial number restoration techniques, the serial number on CTS item 1 (item 2-1-1) was restored to read: D7J274

4WDWEL The attemps of restauration of markings are negatives [sic]
4XR3XH The zinc bar stock (item\#1) was chemically processed. Its serial number was restored to read: D7J274

## Conclusions

672RCF The obliterated area on the piece of zinc bar stock in item 1 was chemically etched and the serial number was determined to be D7J274.

6DFXMD [No Conclusions Reported.]

6DYT6E After application of the electro-acid etching process I determined the serial number of the zinc bar stock as D7J274.

6EYJLV Item: 1 One piece of bar stock described as "a piece of zinc bar stock with suspected obliterated serial number". RESULTS: Item 1 was physically and microscopically examined. The serial number area of Item 1 was prepared and treated with chemical reagents. As a result of these actions, the following characters were recovered D 7 J 274 .

6HCJJM The obliterated serial number has been restored as D7J274.

6KKLGE The primary meaning of the obliterated serial number, most probably, was D7J274
6N3UTK The work on the milled serial nimber[sic] of the bar identified as "item 1 " has given the following result: D7J274

6RNJDX Efforts to restore the defaced serial number produced the characters "D7J274". This should be considered the complete serial number.

6TVVUY Item 1, a piece of zinc bar stock with suspected obliterated serial number was examined within the Firearms Section of the Lab. The suspected area of obliteration was subjected to a series of chemical etchings and the obliterated serial number was fully restored to read: D7J274.

6XBJEW Following a chemical restoration process, I restored the serial number on the Zinc Bar Stock to be "D7J274"

78VZW6 3. On 2016-03-10 during the performance of my official duties I received a sealed evidence bag with number PA6002256159 from Case Administration of the Ballistics Section, containing the following: 3.1 One (1) section of zinc bar with an erased serial number marked by me "70842/16 A". 3.2 One (1) section of an aluminium bar with reference characters (not marked by me). 4. The intention and scope of this forensic examination comprise the following: 4.1 Techniques associated with the recovering and restoration processes of obliterated alpha-numeric figures on metals. 5. I examined the exhibit mentioned in paragraph 3.1 and after application of the electro-acid etching process and comparison with the reference sample mentioned in paragraph 3.2 I determined the serial number on the zinc bar mentioned as D7J274.

7WCH7J The obliterated serial number on the piece of metal (Exhibit 01) was mechanically and chemically treated and restored to read D7J274.

7WPCHD Using standard restoration techniques, the obliterated serial number on item 1 was unable to be restored.

7YFDPM Serial number restoration techniques were applied to item 1 (piece of zinc bar stock). The serial number was determined to be D 7 J 274 .

86Y4M2 Item 001 was examined and found to have an obliterated area. Standard chemical restoration techniques revealed the following characters: "D7J274".

88QYAA Based on the above examination and findings, I am of my opinion that the original serial number on a piece of zinc bar stock "Item 1" is "D7J274".

## Conclusions

8ATXMN Using standard laboratory techniques, the serial number on Item \#1 was restored to read: D 7 J 274

8DEM6F Laboratory examination of the above described piece of evidence reveals the following numbers and or letters to be present D73274.

8GC4GA The restored serial number was $D, 7, J, 2,7,4$.
8XYGXM The Item 1 zinc bar stock was physically and chemically processed in an attempt to restore the obliterated serial number with the following result: The serial number was partially restored to read D7*274 where the asterisk could be either an 8,9 or 0 . The serial number was not searched in any databases.

92WHBP An area of obliteration was observed through the center of Item \#1 (zinc bar) having characteristics consistent with those produced by an end mill type tool. Serial number restoration was successful. The serial number on Item \# 1 (zinc bar) was recovered as: D70274

94H7TG AFTER CHEMICAL PROCESS, THE SERIAL NUMBER REVEALED IS D7J274
96T7RW Item 1-1 one metal-like bar: Visual examination of this item revealed the presence of grind/polish marks on the bar on one side. This area was etched with acid solutions and the following was restored: D 7 J 274

97A3JV The Item \# 1 Bar Stock was physically and chemically processed. Its serial number was restored to read: D7J274.

97NPTX From Proficiency Test \# 16-5250: Item 1: One (1) Zinc Bar Stock with the middle area containing suspected serial number obliterated. Results: Restored serial number for item 1 appears to be : D7J274. Results were verified. Item 1 will be maintained in the firearms evidence storage area along with other completed proficiency tests. I hereby certify that this is a report of the conclusions of an examination performed by me.

99XRPU Item \# 1 was processed using standard serial number restoration techniques and the serial number was restored to read: D7J274

9E3B6M After use of our standard procedures for obliterated serial number restoration we found the following number (left to right): D7J274

9EKJTG The obliterated serial number was restored and concluded to most likely be D7J274.

9FY4LR The serial number was restored to read "D7J274".

9K4Y98 Relvealed[sic] 6 (six) graphic characters removed from the surface of the element.
9KN9QN The serial number has been restored successfully and appeared clearly and entirely.
9L3XU2 As received Item 1 appeared to have the stamped serial number obliterated. Mechanical \& chemical processing was applied and it is the opinion of the examiner that the following characters were developed: D7**74 where the *'s were developed characters however were not discernible enough to determine exactly what they were.

9MVQPK The obliterated serial number of Item \#1 was restored and found to be D7J274.
9N8HJM Using standard laboratory techniques, the serial number on Item 1 was restored to D7J274.

## TABLE 2

## Conclusions

9Q3Y6W Forensic restoration techniques applied to the machined area of the item restored six previously stamped characters that read: D7J274.

9QBD4P Examination of Item 1 revealed one (1) portion of metal bar stock with obliterated serial number. Using standard laboratory restoration techniques, an attempt was made to restore the serial number on Item 1 with the following results: Serial Number: D 7 J 74 was restored on Item 1.

9T3CF3 After application of the electro-acid etching process I could not determine the serial number of the zinc bar stock marked 68097/16A.

9XYRDQ The obliterated serial number was restored to read D7J274.

9Y76NC Upon electrochemical treatment on the field surface, the original number was restored and read as D7J274

A73MD9 Upon electrochemical treatment on the filed surface, the number D7J274 was restored. Based on my findings, I am of the opinion that D7J274 was the original number stamped on the surface that was subsequently obliterated.

A8ZNN7 I am unable to determine the serial number.
AA4JAV Restoration procedures on exhibit SNR1 revealed the serial number to be: D 7 J 274
ACA28Q Exhibit 1 (Item 1) - One piece of Zinc bar with an obliterated serial number. The serial number was chemically restored to be: D7J274.

ADMNRQ The Item \# 1 metal alloy bar displayed an apparent area of obliteration as received. The serial number was restored and found to be D7J274. The metal alloy bar with the stampings was not examined.

AEYADL Restoration of obliterated stamped markings was performed on the questioned surface of ltem 1 and the original serial number was found to be "D7J274".

AJDY2N I restored the serial number for item 1 to D7J274.
AJM3G9 Using standard laboratory restoration techniques, the obliterated serial number was restored to read: D 7 J 274

AJRJE7 The above number was obliterated through mechanically[sic] obliteration of metal surface from the serial number field.

ALYRUH Serial number restoration techniques were applied to the submitted zinc bar stock (item \#1). The serial number was determined to be D7J274.

AVGH79 The bar received is not that of a zinc, it looks like an aluminium. The number (serial) was never punched or it was removed during obliteration. [From Table 1 - Recovered Characters: "NOT FOUND"]

AXBZT4 The following serial number: D7J274 was recovered and read out as the result of examination of the item designated as Item 1. Decoded alphanumeric characters were compared with the characters located on sample designated as standard. The standard sample was made of aluminium, whereas Item 1 was paramagnetic.

B284XV The obliterated serial number, located in the middle of the zinc bar stock nearest to the right side, was chemically processed. Attempts to restore the obliterated serial number were successful. The restored serial number is "D7J274". The serial number was verified by [Analyst] of the [Laboratory].

## Conclusions

B4ELVR Defaced serial number was chemically restored to "D7J274".
B67GKL Standard laboratory techniques for restoring effaced markings in metal have been employed on the inset portion of the zinc bar stock. The serial number was restored and observed to be D 7 J 274.

BAHHPP Serial Number Restoration Analysis: Methodology - Chemical reagent Etching / Microscopy / Physical. Serial number restoration procedures revealed the serial number on Item 1, the zinc bar stock, to be: D7J274.

BHFXRQ Using standard laboratory restoration techniques, the obliterated serial number on Item \#1 was restored to read D7J274.

BWCAZU The obliterated serial number on the piece of bar stock (Item 01), was chemically restored and determined to be "D7J274".

BWXQFX Visual examination and chemical treatment of the serial number area on the bar stock, Item 1A, reveal the following number: D7J274. Item 1B was inspected to verify and document contents. No analysis was performed on the item listed.

BZCKVU The serial number of Item 1 was restored to read D7J274.
C4T88E Examination and restoration of the obliterated area on Item 1 (a piece of zinc bar stock with suspected obliterated serial number) revealed the following characters interpreted as "D7J274".

C4VPEP A piece of metal was received at the laboratory that had a ground area on one side. Number restoration was attempted on this area using chemical etching methods. The following characters were recovered: "D 7 J 27 4"

C6Q9FR Serial Number Restoration Analysis: Methodology - Chemical Reagent Etching / Microscopy / Physical. Serial Number restoration procedures revealed the serial number on Item 1, the Zinc bar stock, to be: D7J274.

CAGUZY 3. On 2016-03-08 during the performance of my official duties I received a sealed evidence bag with number PA6002256160 from Case Administration of the Ballistics Section, containing the following:
3.1 One (1) sealed envelope marked "Test No. 16-5250: SERIAL NUMBER RESTORATION", containing the following: 3.1.1 One (1) envelope marked "Test No. 16-5250 Item 1 ZINC BAR STOCK", containing the following exhibit: 3.1.1.1 One (1) piece of metal marked by me "70828/16A". 3.1.2 One (1) envelope marked "Aluminum Standard Test No. 16-5250", containing the following: 3.1.2.1 One (1) piece of metal with numbers " 0 to 9 " and letters "A to F" and "H", "J", "K" and " N ", not marked by me. 4. The intention and scope of this forensic examination comprise of the following: 4.1 Techniques associated with the recovering and restoration of obliterated alpha-numeric figures on metals. 5. After application of the electro-acid etching process, I determined the number on the piece of metal mentioned in paragraph 3.1.1.1 as D7J274.

CAV438 The serial No.area of the piece of zinc bar was prepared and chemically processed with restoration reagents. As a result of these actions the serial No. was restored to read D7J274

CCKPJ2 The serial number was fully restored to read "D7J274".
CGG7YG One (1) piece of zinc bar stock (approx $21 / 4^{\prime \prime} \times 1^{\prime \prime}$ ) submitted with a suspected obliterated serial number. A $11 / 4^{\prime \prime} \times 1$ "rectangular area of surface was removed by grinding. Serial number "D7J274" restored using chemical etching process. Bar stock was scribed with "CTS 16-5250" examiner.

CNGTNV Visual examination and chemical treatment of the serial number area on the zinc bar stock, Item 1A, reveal the following number: D7J274.

## TABLE 2

## WebCode

## Conclusions

CYJQW6 The serial number of Item 1 was restored and read as D7J274
CZPHF6 The obliterated serial number located on the center portion of the zinc bar stock was chemically processed and restored to read "D7J274".

D8MK7N Using standard laboratory techniques, the obliterated serial number on Item 1 was restored to read "D7J274."

DDABPX After application of the etching process, I determined the serial number on the zink[sic] bar stock as D7J274.

DFFTQE The item is a zinc bar stock with suspected obliterated serial number. Laboratory chemical restoration procedures revealed the following serial number: D7J274.

DGXMRN Using standard laboratory restoration techniques, the obliterated serial number on ltem 1 was restored to read D 7 J 274 .

DHNT8E The serial number has been restored to read: D7J274.
DRF6JV Attempts to restore the serial number were unsuccessful.
DTLUKJ The serial number of the metal plate was restored and determined to be D7J274.
DVV2MC Serial number restoration was preformed[sic] on item 1.1. The serial number D7J274 was restored.
E2WBVP The Item \# 1 was physically and chemically processed. Its serial number was restored to read D7J274.
E6Y96W After application of the etching process, I determined the serial NR on the zinc stock bar as D7J274.
E77NDN The serial number was chemically processed and restored to read D7J274
E7QG4Z The serial number of the zinc bar stock, Item SNR1, was determined to be D7J274.
E9WUUE Chemicals[sic] solutions were applied to the altered serial number. The following number has been revealed: D7J274. See attached picture. [Picture is unable to be included in this report.]

EHZ63P Examination and processing of the Q-1 bar stock restored the original obliterated serial number, which was determined to be D7J274.

EJTXX9 THE METAL PIECE RECEIVED FOR RESTORATION OF SERIAL NUMBERS WAS REVELATED BY THE CHEMICAL REVALING METHOD (FRY); THE SERIAL NUMBER REVELATED WAS D7J274.

EJYBJJ The obliterated serial number on the Item 1 zinc plate was restored to read D7J274.
EP43WM The serial number was successfully restored.
EQU347 1. The obliterated area on Exhibit 1 (piece of zinc bar stock) was visually examined, polished, and chemically processed. The characters were restored and appeared as follows: D7J274.

ERB8L6 After preparation and treatment of the serial number area of Item 1, no clearly identifiable serial number was developed.

ERPVPH Standard laboratory techniques for the restoration of effaced stampings in metal were employed on the Exhibit 1 zinc bar stock. The serial number was restored to read D 7 J 274.

## Conclusions

EUFYDH Using standard serial number restoration techniques, the obliterated serial number on item 1-1-1 was restored to read: D7J274.

F33DUT Standard restoration techniques were applied to Item 1. The following characters were restored: D7J274.

F4JCCX In the piece of metal was restored the alphanumeric sequence D7J274
F8VGLC The Exhibit \# Item 1 obliterated serial number was restored to read "D7J274".
F9QZMD ITEM 1: ONE PIECE OF ZINC BAR STOCK WITH A SUSPECTED OBLITERATED SERIAL NUMBER, MARKED K1. Q1 SERIAL NUMBER "D7J274" WAS FULLY RESTORED USING POLISHING AND ACID ETCHING SOLUTIONS.

F9TPUW After application of the electro acid etching process, I determined the serial number of the exhibit mentioned in 3.1 as D7J274.

FM9WK2 The serial number of Item 1 was restored using chemical etching techniques and was found to be: D7J274.

FR6CHP Exhibit 1 has an area of obliteration in the center of the zinc bar stock. The obliteration appears to be accomplished by milling and the resulting toolmarks are suitable for identification. The area on the zinc bar stock was chemically processed and the serial number was fully restored to read D7J274.

FTX96X The serial number of the zinc bar stock, described in the item 1, was restored and corresponds to: D7J274.

FV4TTX The obliterated serial number was fully restored to read "D7J274".

FYLT9D I found filing marks on the metal plate Item 1. Upon electrochemical treatment on the filed surface, the number 'D73274' was restored. Therefore, I am of the opinion that the obliterated serial number is D73274.

G7GCPZ The grinded surface on the zinc bar stock was electrochemically treated and a set of alphanumeric was restored and read as "D7J274"

G88KP3 After application of the electro-acid etching process, I determined the serial number of the exhibit possibly as D7J274.

GA4PQQ The "Item 1" obliterated serial number, located to the right/center of the zinc bar, was chemically processed and partially restored to read "* $7 * 274$ ", where the first asterisk most likely represents a " 0 " but a "D" could not be eliminated, and the second asterisk most likely represents a "0" but a "J" or a "9" could not be eliminated.

GBXHLA Using chemical etching techniques the serial number was restored to read D7J274. The aluminum standard was not further examined.

GCAAGC Serial number restoration techniques were applied to Item 1 (bar stock). The serial number was determined to be D7J274.

GF7XVX he[sic] serial number "D7J274" was recovered from the piece of zinc bar stock, identified Item 1.
GG8LKN The metal pad displayed grind marks where the serial number is usually stamped. I chemically restored the digits D7J274 which appears to be consistent with the supplied reference sample digits.

| WebCode | Conclusions |
| :---: | :---: |
| GJAW7Q | Visual examination and chemical treatment of the serial number area on the bar stock, Item 1, fail to reveal the original serial number. |
| GLDFEK | Serial number restoration analysis: Serial number procedures revealed the serial number on Item SNR1A, the Zinc bar to be: D7J274. Methodology: chemical reagent etching / microscopy / physical. |
| GUMWZY | THE BAR HAS BEEN GRINDED. AFTER THE ANALYSIS WAS DONE ON THE GRINDED SURFACE, THE NUMBER WAS RESTORED AND READ AS D7?274 (WHERE "?" REPRESENT THE NUMBER OR ALPHABET THAT CANNOT BE IDENTIFIED DUE TO INTENSIVE GRINDING). |
| GVZPED | Examination and chemical processing of the obliterated serial number in Item \#1 restored the following sequence: "D 7 J 27 4". Item \#2 was for reference purposes only. |
| GWW7QB | The serial number was restored to read D7J274. |
| GWYJMJ | Physical and chemical processing of the submitted bar stock restored the obliterated, original serial number to read "D7J274". |
| H24RGT | 3. On 2016-3-10 during the performance of my official duties I received a sealed evidence bag with number PA6002256149 from Case Administration of the Ballistics Section, containing the following: 3.1 One (1) inner CTS-Evidence sealed brown envelope, containing the following: 3.1.1 One (1) brown tape-sealed envelope, marked "Aluminum Standard" and "Test No. 16-5250", containing the following item: 3.1.1.1 One (1) Aluminum bar stock intended as a Standard, with the figures "1" to "9" and "O" and the letters, "A" to "F", "H", "J", "K" and "N" there on, marked by me with "76121/16 Al.St". 3.2.1 One (1) brown tape-sealed envelope, marked Test No. 16-5250" and "Item 1", containing the following exhibit: 3.2.1.1 One (1) piece of zinc bar stock, marked by me with "76121/16 Item 1". 4. The intention and scope of this forensic examination comprise of the following: 4.1 Techniques associated with the recovering and restoration processes of obliterated alpha-numeric figures on metals. 5. After the application of the electro-acid process on the exhibit as mentioned in paragraph 3.2.1.1, I determined the obliterated alpha-numeric figures as D70274. |
| H7CC36 | THE ZINC BAR STOCK RECEIVED FOR RESTORATION OF SERIAL NUMBERS, WAS REVEALED BY THE CHEMICAL REVELAING[sic] METHOD (PALMERTON); THE SERIAL NUMBER REVEALED WAS D7J274. |
| H8BHJY | On the examination, I found that there were filing mark on the surface of the zinc bar stock and no numbers were observed. On electrochemical treatment, a set of number "D7J274" was restored. Hence, I am of the opinion that the numbers of the zinc bar stock were tempered and the original number numbers[sic] were "D7J274". |
| HC7ATG | The obliterated serial number on the zinc bar, Item 1, was restored to D7J274. |
| HX6B7K | Using standard laboratory restoration techniques the obliterated serial number on Item 1 was restored to read: D7J274 |
| J46GW8 | The obliterated six alphanumeric characters on the zinc bar stock were restored as followed: D 7 J 27 4 |
| J4M4UN | The obliterated area on Exhibit 001 (metal bar stock) was visually examined, polished, and chemically processed. The characters were restored and appeared as follows: D 7 J 274. |
| J6ZZX9 | The serial number was restored to read "D7J274". |
| J7FAYL | The zinc bar stock had an area of material removed using a mill type tool. The milled area was treated with a $10 \%$ nitric acid solution and the following characters were restored: D7J274. |

## Conclusions

JEL2F4 The serial number on the piece of zinc bar stock was restored to read D7J274.
JEQCKM The area of metal was chemically examined and the characters D7**74 were recovered. The '?' represents an unknown character. Fragments of these characters were recovered, however were of insufficient clarity for a positive identification.

JGQNVT The serial number observed on the zinc bar was D7?274, the simbol[sic] "?" is a character that was not possible to restore

JKEZWJ The serial number was chemically processed and restored to read "D7J274".
JKQZQT The restoration of the metal piece was realized and was achieved to reveal the alphanumeric sequence D7J274

JLLHTU The recovered number was found D7J274, in accordance with aluminium pattern.
JNECRJ The serial number was chemically restored to D7J274.

JP7B4W Attempts to restore the serial number on item \#1 with polishing and chemical processing successfully restored the serial number to read, "D7J274".

JRXGBF Results: The restored serial number was: D73274. *Note: It is the responsibility of the case agent to research the restored serial number and enter the restored serial number in the Property Evidence Tracking System (P.E.T.S).

JT99GB One (1) block of silver non ferrous metal measuring approximately $21 / 4^{\prime \prime} \times 7 / 8^{\prime \prime} \times 1 / 8^{\prime \prime}$ displaying a 1 1/4" x 1" area milled away - serial number D7J274 recovered with chemical etching. Item marked 16-5250 for identification.

JYWTKQ The submited piece of zinc barstock, with a suspected obliterated - serial number was examined. An attempt was made to recover the serial no. The serial no. D7J274 was recovered.

K3K6MZ THE SERIAL NUMBER HAD BEEN ERASED; HOWEVER I RESTORED IT BY MEANS OF CHEMICAL ETCHING. THE SERIAL NUMBER WAS D7J274.

K4KTEC Item 1 is a piece of stainless steel colored metal bearing an obliterated area which was restored to read D7J274. The serial number in Item 1 was restored using polishing tools and various chemical etchants.

K4ZEND The obliterated serial number on the item 1 piece of zinc bar stock was restored and found to be D7J274.

K7PK44 AFTER PERFORMING THE SERIAL NUMBER RESTORATION TEST OVER THE PROVIDE ZINC BAR THE FOLLOWIN[sic] SERIAL NUMBER WAS RECOVERED: D7J274.

K9V7WB Standard laboratory procedures for restoring characters stamped in metal have been employed on the recessed area of the submitted bar stock. The serial number was determined to be "D7J274".

KALAKB Chemical restoration of number on piece of zinc bar start time : 15h00/finish time 15h15. Number stamped: D7J274

KCV792 After application of the chemical process, I determined the serial number of item SNR1 to be D7J274.
KD8UR2 The firearm, item \# 1, was chemically processed. Serial number restored to read (D7J274).

## Conclusions

KRT9VZ Examination and chemical processing of Item 1 determined the original obliterated serial number to be D7J274.

KXFWQ7 The defaced serial number on the piece of zinc bar stock (Item 1) was restored to be "D7J274".
KZLGFQ After application of the electro acid etching process, I could not determine the alpha numeric figures on the zinc bar (mentioned in 3.1 of statement).

L36WLC The serial number of Item 1 was mechanically and chemically processed and restored to read "D7J274".

L4HKM4 The serial number on Item 1 has beenobliterated[sic] by a milling process. Our attempt to restore this numbre[sic] was successful, revealing the following characters: D 7 J 274

LG4ZQ4 The serial number on the piece of metal (Exhibit 1) was obliterated through the middle of the numbers as submitted. The obliterated serial number was mechanically and chemically restored on the piece of metal (Exhibit 1) to be D7J274.

LGNLG9 ONE (1) ZINC BAR STOCK WAS RECEIVED WITH AN OBLITERATED SERIAL NUMBER. SERIAL NUMBER OF D7J274 FULLY RESTORED BY POLISHING AND CHEMICAL MEANS.

LGPBMQ After application of the electro-acid etching process I determined the serial of the item 1 as D7J274.
LHVXXA The obliterated serial number of Item 1 was restored to be D7J274.
LJQGZB Item 1: The serial number was restored to read D7J274.
LN3JQD The serial number had been obliterated from the Zinc bar and was restored with light sanding and chemical etching to yield the serial number D7J274.

LQD8M7 The serial number was restored to read: D 7 J 27 4. The aluminum bar stock was not further examined.

LV4JLA The obliterated serial number on the section of zinc bar stock in Item \#1 was completely restored and found to be D 7 J 274.

LV92K8 The obliterated serial number of the Item \# l zinc bar stock was chemically restored to read D7J274.
LVJZG3 THE IDENTIFICATION NUMBER OF THE PLATE CORRESPONDING TO TEST N ${ }^{\circ} 16-5250$ ITEM 1 (ZINC BAR STOCK) REVEALED THE FOLLOWING ALPHANUMERIC CHARACTERS "D7J274" BEING DETERMINED TO MATCH THE TYPE OF MARKING STAMPED ON THE PLATE IDENTIFIED AS ALUMINUM STANDARD TEST N ${ }^{\circ}$ 16-5250

LVPC3D The obliterated serial number on the zinc bar stock, item 1, was restored to D7J274. The zinc bar stock, item 1, was examined noting that the location of the suspected serial number was obliterated by a grinding type of tool. Using standard restoration techniques, the area was sanded and treated with chemicals.

M4TPUZ The above procedeure[sic] revealed the following alphanumeric characters "D7J274" corresponding to the identification number of the metal plate described in the EMP 1 ID (item 1), these match the dialing mode used by the manufacturer Collaborative Testing Services.

M4Y2FC The chemically enhanced serial number on item 1 is D7J274.

## TABLE 2

## WebCode

## Conclusions

M6P8W3 The zinc metal bar stock (1) was physically and chemically processed. Its serial number was fully restored to read: D7J274 on 04/05/2016.

M73WYE Item 1 was physically and microscopically examined. The serial number area of Item 1 was prepared and treated with chemical reagents. As a result of these actions, the serial number was successfully restored to read "D7J274".

M8UVBR The piece of metal has surface wear by chemical restoration the original sequence (D7J274) was restored, which corresponds to the pattern characters used as comparators.

MD7UTC 1. Exhibit 1 is a zinc bar stock with an obliterated serial number. 2. Mechanical and Chemical processing of Exhibit 1 revealed that the original serial number is D 7 J 274.

MGHQB8 The examination and processing of the obliterated serial number on the ltem 1 piece of bar stock was restored to read "D7J274".

MMNK64 THE ITEM 1 OBLITERATED SERIAL NUMBER, LOCATED ON THE FRONT MIDDLE OF THE ZINC BAR STOCK WAS POLISHED USING THE DREMEL AND CHEMICALLY PROCESSED AND RESTORED TO READ D7J274.

MRK49M After application of the electro-acid etching process, I dermined[sic] the serial number of the bar stock as D7J274.

MU9FAD One side of Exhibit 1 has a portion where they[sic] metal appears to have been removed using a milling machine. This area was polished and chemically processed. The serial number D 7 J 274 was restored on Exhibit 1.

MVZCY8 Serial number on MBI (metal block) was removed (deeply abraded). The serial number was restored using the chemical etching process. CTS number etched on MBI for identification.

MWYLQ4 THE SUBMITTED BAR STOCK WITH THE OBLITERATED SERIAL NUMBER WAS MARKED Q1 FOR IDENTIFICATION. Q1 WAS POLISHED AND CHEMICALLY PROCESSED. THE SERIAL NUMBER RESTORATION REVEALED THE FOLLOWING D 7 J 274.

MZYPKC The serial number of Item 1 as restored is D7J274.
N6MWFP The serial number of the zinc bar stock described in the item 1, was restored and corresponds to: D7J274.

N7HFGQ Alphanumeric sequence "D7J274" was restored in the disturbed area of the object identified as "item 1"
NF84EE Item 1 is a piece of zinc bar stock with an obliterated serial number. Using standard serial number techniques, the serial number on Item 1 was restored to read: D 7 J 274.

NFJZJJ Visual examination and chemical treatment of the serial number area on the zinc bar stock, Item 1, reveal the following number: D7J274

NMGRQR The suspected obliterated serial number was restored and found to consist of the following six (6) characters D 7 J 274.

NQ64RH Serial number restoration was conducted on item 1, the Zinc bar stock, using acid etching chemical processing techniques. The following characters were revealed: *7*274. The first * was not clear. It appeared to be "D", but could be also " 0 " (zero). The second * was not clear. It appeared to be a "0"(zero) or "8", but could also be another character.

## TABLE 2

## WebCode

## Conclusions

NRW34V Attempts to restore the obliterated serial number of Item 1 were successful. The restored serial number is D7J274.

NTUDYY The obliterated area on the piece of zinc bar stock in item 1 was chemically etched and the serial number was determined to be D7J274.

NWTRK3 Using chemical etching techniques the serial number was restored to read D7J274.
NX7GP2 The serial number on the piece of metal (Exhibit 01) was mechanically and chemically treated and restored to read D7J274.

PAF8BA The obliterated area was partially restored to reveal a six character serial number. The third character appeared to be a " J " but a " 0 " could not be ruled out. Consequently the possible serial number combinations are, "D7J274" and "D70274".

PGC969 The serial number of Item 1 was mechanically and chemically processed and restored to read "D7J274". This conclusion was verified by Firearms Examiner $\qquad$ .

PJXXMZ In the piece of zinc BAR STOCK with suspected obliterated erial[sic] number, that comes identified as " test $N^{\circ} 16-5250 "$ item 1 . Submitted to the procedeure of developing serial number, the following alphanumeric characters was obtained D73274.

PL4HDK After application of the electro-acid etch process I determined the serial number on the zinc bar stock mentioned in paragraph 3.1 as containing six (6) alpha - numeric characters and is recorded as D7J274.

PP8ZPV The numbers were unable to be restored.
PWLY6R On analysis, I found there was a filling mark on the surface of the steel bar. On electrochemical treatment on the filled surface region, I found a number "D7? 274 " where ? is the undermined alphabet / number due to intensed filled on the metal surface area. Hence, I am of the opinion that the number of the steel bar was tempered and the original number was D7? 274 .

PZKB4D Examination and processing of the Q-1 bar stock restored the original obliterated serial number which was determined to be D7J274.

Q2A4B2 The serial number was restored to read D70274.
Q7TYET The obliterated serial number was restored to read D7J274
Q9XLV4 The Exhibit's surface was lightly polished, using grinding paper 600. The polished surface was then treated with HF based solution. The restored serial number is D7j274. The results were successfully photographed.

QA9NH3 Serial number restored using chemical etching process with the following results: "D7J274" FIU \# scribed on evidence by examer[sic] for identification purposes.

QAT6X7 Standard laboratory procedures for restoring serial numbers stamped in metal have been employed on the cut out of this piece of metal. The serial number was restored to : D7J274.

QD9ZD3 Serial number obliterated by abrasion. Serial number restored by chemical etching and reads "D7J274".

QLYKGF Examination and chemical processing of the zinc bar failed to restore a complete serial number restoration.

## Conclusions

QRPVRE THE OBLITERATED SERIAL NUMBER WAS RESTORED TO READ "D7J274".
QUE4YU Item (1) was physically / chemically processed. It's serial number was restored to read (D7J274).
QY82Q7 Standard laboratory techniques for the restoration of effaced stampings in metal were employed on the section of zinc alloy Exhibit 1. The serial number was restored to read: D 7 J 274.

R6TFRH 3. On 2016-03-30 during the performance of my official duties I received a sealed evidence bag with number PA6002256158 from Case Administration of the Ballistics Section, containing the following:
3.1 One (1) $57 \mathrm{~mm} \times 23 \mathrm{~mm}$ piece of Zink[sic] bar of which the serial number is erased, marked by me " $78729 / 16$ A". 3.2 One (1) $50 \mathrm{~mm} \times 38 \mathrm{~mm}$ piece of Aluminium bar sample set with punched numbers as follows: "A" to " N " and also " 0 " to "9" respectively, marked by me "78729/16 B". 4. The intention and scope of this forensic examination comprise of the following: 4.1 Techniques associated with the recovering and restoration processes of obliterated alpha-numeric figures on metals. 5. After application of the electro-acid etching processes I determine the serial number of the piece of Zink[sic] bar mentioned in paragraph 3.1 as D7J274.

RD7CU4 Restoration of the obliterated serial number was performed on the questioned surface of the zinc bar stock marked "ltem 1". The restored serial number was found to have six characters - "D7J274".

RDLTPW Item (1) was physically / chemically processed. It's serial number was restored to read: D7J274.
RKPVUW The obliterated characters on the piece of zinc bar stock (Exhibit 1) were chemically and mechanically treated and were restored to read "D7J274".

RL43FE Attempts to restore the obliterated serial number on Item 1 were successful. The obliterated serial number was restored to read "D7J274".

RNQRLD The area of the obliterated number was polished and chemical etchants then applied. This resulted in the digits: D7J274 being restored.

RPG3C7 The serial number on the zinc bar, item SNR1-1, was partially restored to read "07*274". The "*" could be either a "0", "3", "6", "8" or a "J".

RR2QUX The serial number was removed approximately 0.2 mm thickness. The removed serial number was resulted by the examination: D73274.

RR79TW The piece of stainless steel bar stock sent to study presents six characters printed as follws[sic]: D7J274 for testing frys.

RZWX6Q Results of Laboratory Examination: Examination and restoration of the obliterated area on Item 1 revealed the following characters "D7J274".

T6PDDX Using chemical etching techniques, the serial number was restored to read D 7 J 274.
TC8L4F Item \#1 was received with an obliterated serial number. Chemical etchants were applied to the obliterated area. The characteristics were restored to read D7? 274 , where "?" is likely a J or 0.

TKZ48P After application of electro-acid etching process, I could not determine the serial number on zinc bar stock. [From Table 1 - Recovered Characters: "Not found."]

TQ7MH2 The serial number on item 1A was restored to read D7J274

## Conclusions

TXN2G2 THE ZINC BAR STOCK (ITEM 1) WAS MARKED Q1, [Name], [Participant Code] TO THE LEFT OF THE SUSPECTED OBLITERATED SERIAL NUMBER AREA. THE AREA WITH THE SUSPECTED OBLITERATED SERIAL NUMBER WAS POLISHED WITH A DREMEL TOOL, AND THEN THE CHEMICAL ETCHING SOLUTIONS WERE APPLIED. THE OBLITERATED SERIAL NUMBER WAS PARTIALLY RESTORED TO READ AS FOLLOWS: ?? ? 274.

TXQQNJ After application of the electro etching process, I determined the serial number on the zinc bar stock as D7J274.

U2YFFQ The submitted sample is a piece of zinc bar stock with a milled down area on one side. The sample was sanded down in the milled area, using grit size 800 sanding paper until an even surface was achieved and then etched using Fry's solution. The obliterated serial number was restored with this method and was clearly visible.

U8BEXC The serial number is milled off. The serial number (D7J274) was restored by acid etching. The modified Fry's Reagent and the Nickels \& Alloy's Reagent were used.

UDWAYF I have applied electro-acid etching method, I determined the serial number of the item (1) as D7J274.
UJ34VW The serial number on the piece of metal (Exhibit 2) was mechanically and chemically treated and restored to read D7J274.

UP7YQP After the application of the electro-magnetic etching process, I could not recover the serial number of the exhibit mentioned in 3.1 (marked 74618/16 A).

UVWA2N Using standard laboratory restoration techniques, the serial number on Item 001A was restored to read D7J274.

UYWKWN Using standard laboratory restoration techniques, the serial number on the piece of aluminum in item 001A was restored to read D7J274. The piece of aluminum in item 001B was submitted for continuity.

UZNR37 Visual examination and mechanical and chemical processing of the piece of bar stock (ltem 1) revealed the serial number to be D7J274.

V62RZX Characters "D7J274" restored by chemical etching.
VE976K I found the serial number to have been obliterated by filing. On electrochemical treatment, I developed the number D7J274. I am of the opinion that the original serial number was D7J274.

VJZQV8 The acid etch method was used to completely restore the serial number of the 0001-AA (Item 1) Zinc bar stock to read: D7J274. No testing was conducted on or with the 0001-AB Aluminum standard.

VM3CZ8 Using standard laboratory restoration techniques, the obliterated serial number on Item 1 was partially restored to read D7*274, where * designates a non-recovered character.

VP6ABZ The obliterated serial number on Item 1 was restored and interpreted as "D7J274".

VRA33D The obliterated serial number on the Item 1 bar stock was restored to read: D7J274.

VYBGNE After application of the electro acid etching process with green mamba acid I determined the serial number of the exhibit (zinc bar stock) item 1 as D7J274.

VZ68UT The obliterated surface on the Zinc bar stock (ltem 1) was sanded and chemically processed. All characters could have been seen almost in same time during the examination.

## Conclusions

W2F8QM The obliterated serial number on the zinc bar stock (item 1) was restored to read D7J274.
W4LRF9 The serial number on the piece of zinc bar stock was obliterated by a milling process. No tool marks of value were observed. The obliterated area was polished and chemically processed. The serial number "D7J274" was observed.

W8L3AM Examination of Item 1 revealed the presence of a defaced area. Item 1 was physically and chemically processed. The serial number was restored as: D7J274

WBJJMG Chemical treatment was successful in chemically restoring a serial number on the bar. The serial number on the bar was restored to read D 7 J 274.

WCARLK The obliterated serial number stamped on Item 1 was chemically processed and restored to read "D7J274."

WD8VN9 \#1, the obliterated serial number on the piece of zinc bar stock was unable to be restored.
WFA3HK 1. The obliterated area on Exhibit 1 (metal block) was visually examined, polished and chemically processed. The characters were restored to read: D 7 J 274.

WGAUYZ The obliterated number on Item 1 was chemically restored to reveal the serial number D7J274. Supporting examination documentation is maintained in the case file.

WL22EN Visual examination of the one (1) piece of zinc bar stock, item \#1, revealed an obliterated area on its surface. Chemical etching of this area revealed the serial number to be D7J274. The additional submitted one (1) aluminum standard, item \#1, was not examined at this time.

WL79U3 A serial number restoration was performed on Laboratory Item 001.A Zinc bar stock. Based upon CTS, the expected serial number configuration for Laboratory Item 001 . A Zinc bar stock is six characters. The serial number was fully restored and appeared to be D7J274.

WQWTP7 Examination and processing of the Q-1 bar stock restored the original obliterated serial number, which was determined to be D7J274. A search of the NCIC Stolen Gun Files by the restored Serial Number was not done as this is a proficiency.

WRRBR8 Serial number restoration reveals the number "D7J274".
WTK8DF After application of the electro acid etching process, I determined the serial number of the exhibit mentioned in item 1 (3.1) as D7J274.

WTMY7W The Exhibit \# 1 serial number was restored to read D7J274.
WXJHC2 The obliterated serial number on the metal bar, item 1-1, was restored to read "D7J274".

XA4J3R The serial number was restored to read: D7J274.
XNM3Z3 Examinations showed the serial number of Item 1 to be obliterated. The serial number of Item 1 was restored through chemical etching techniques and was found to read: D7J274.

XQEYNW One (1) piece of zinc bar stock (approx $21 / 4^{\prime \prime} \times 1^{\prime \prime}$ ) submitted with a suspected obliterated serial \#. Surface of the material is defaced by concentric abrasions. Serial \# "D7J274" restored using chemical etching process, scribed with number "16-5250" by examiner for identification.

Y8DM7L After application of the electro-acid etching process, I could not determine the serial number of the zinc bar stock.

## TABLE 2

## WebCode

## Conclusions

Y96TC4 The serial number is milled off. The serial number (D7J274) was restored by acid etching. The modified Fry's reagent and the Nickels \& Alloys reagent were used.

YBWP2X The section of aluminum bar stock Ex. 1.1 was visually inspected. Standard Laboratory techniques for the restoration of effaced markings in metal were applied to the damaged area of the section of zinc bar stock Ex. 1. The characters appeared to be D 7 J 274.

YKDLTH Examination and chemical processing of Item \#1 restored the following characters "D7J274".
YKNYU2 Mechanical and chemical processing of the submitted exhibit 1 revealed that the original serial number is D7J274.

YMPC8T Item 1 was physically, chemically and magnetically processed. Its serial number was restored to read: "D 7 J 27 4"

YXBFER The serial number on the piece of metal (Exhibit 1) was obliterated through the middle of the numbers as submitted. The obliterated serial number was mechanically and chemically restored on the piece of metal (Exhibit 1) to be D7J274.

Z9JKYT The obliterated serial number of Item \# 1 was restored to D7J274.
ZAXZG2 Methodology - Chemical Reagent Etching / Microscopy / Physical. Serial number restoration procedures revealed the serial number on Item 1, the bar stock, to be: D7J274.

ZDHU9V Using standard laboratory techniques, the serial number/identification data on ltem 1 was restored to: "D 7 J 27 4"

ZE9ZPL The zinc bar stock (item\#1) was chemically processed. Its serial number was fully restored to read D7J274.

ZJMXW9 Examination of Item 1 revealed an obliterated area on one side. Standard chemical restoration techniques revealed the following characters: "D7J274".

ZL47ZE Test No. 16-5250 Item 1 ZINC BAR STOCK is represented in a paper package. The testing material provided is a piece of zinc bar stock, according to the note on the package. The aim of the test is to restore the deleted serial number. Unfortunately, criminalistics examination department of [Laboratory] doesn't have relevant zinc material processing reagents, which would assist us to complete the serial number restoration. After the examination of the zinc bar stock under the violet rays $395-425 \mathrm{~nm}$ of "Crime-lite M2", using GG455-AG filter and VSC 6000 / HS ultraviolet rays 365 nm , polishing it with emery cloth and processing it with the chemical reagents available to us, the serial number restoration was not possible.

ZX3V4U Using standardized laboratory techniques, the serial number of bar stock Item 1 was restored to read: D7J274.

ZYVNYD The serial number of the zinc bar stock described in item 1, was restored and corresponds to: D7J274.
ZZWCN4 A series of previously stamped characters were restored which read: D7J274. The size and font were similar to those used on the aluminium bar stock labelled "standard".

## Sample Preparation

## (listed in order of use)

TABLE 3

| WebCode | Method | Tool Used | Grit Size |
| :---: | :---: | :---: | :---: |
| 26JQKN | Visual | Stereoscope |  |
|  | Polishing | Rotary Tool | polishing wheel |
| 27EALP | Polishing | Rotary Tool |  |
| 2ACVB8 | Visual |  |  |
|  | Polishing | Dremel | fine |
| 2B8DD9 | Sanding | Dremel Europe | 1 mm |
|  | Polishing |  |  |
|  | Visual |  |  |
| 2BNLMW | Sanding | Dremel Tool \& Sandpaper | $150+320$ grit |
| 2CJ4NX | Polishing | Rotary Tool |  |
| 2NTQPH | Sanding | Sand paper | Wet/Dry |
| 2P7FRV | Clean \& Micro. Exam | Acetone (clean) |  |
| 2QJRWR | Polishing | Dremel |  |
| 2QXCGP | Buffing/Polishing | Dremel w/ polish wheel |  |
| 2QZ9ME | Sanding | Dremel |  |
| 2WKBKH | Sanding | Sand paper | 280 |
|  | Polishing | Sand paper | 400 |
| 2XW2XT | Polishing | Dremel |  |
| 2ZQHH3 | Visual | My eyes |  |
|  | Cleaning | Prepsol |  |
|  | None | Cast surface for file |  |
|  | Sanding | Emery paper | 1200 grit |
|  | None | M7 V1-Use posi-test |  |
| 32BX2A | Sanding | Sand paper | 400, 220 and 80 |
| 33R6BX | Grinding | Dremel |  |
|  | Polishing | Dremel |  |
| 34923 x | Grinding | Dremel |  |
|  | Polishing | Polishing wheel |  |
| 3836 JH | Sanding | Sand paper | MED |
| 3B64WW | N/A | N/A | N/A |

## TABLE 3

| WebCode | Method | Tool Used | Grit Size |
| :---: | :---: | :---: | :---: |
| 3BZQBL | Visual | Stereoscope |  |
|  | Polishing | Sand paper | 150 then 200 |
|  | Visual | Stereoscope |  |
| 3D6C27 | Polishing | Dremel |  |
| 3MUZXU | Polishing | Dremel tool |  |
| 3NPHZV | None |  |  |
| 3PK7EK | Polishing | Rotary Tool |  |
|  | Cleaning | Sand paper | 600 |
| 3QFPFL | Cleaning | Acetone |  |
|  | Sanding | Sand paper | 800 |
|  | Polishing | Steel wool |  |
| 3RTGBN | Visual | magnification goggles |  |
|  | Grinding | Dremel |  |
|  | Polishing | Dremel |  |
|  | Cleaning | methanol |  |
|  | Visual | magnification goggles |  |
| 3UD7U2 | Sanding | Sand paper | 120 |
| 3ZJYPW | Polishing | Dremel | fine grit |
| 438NVV | Sanding | Emery paper | Medium |
|  | Polishing | Sand paper | 600 grit |
| 477ZQV | visual w/ oblique light | Flashlight |  |
|  | Polishing | Dremel tool |  |
| 482HTW | Polishing | Dremel |  |
| 4GK94M | Visual | Stereoscope |  |
|  | Visual | digital camera |  |
|  | Sanding | Sand paper | 400 |
| 4LE34A | Visual | Stereoscope | N/A |
|  | Cleaning | Acetone | N/A |
|  | Polishing | Sand paper | P220 |
|  | Polishing | Sand paper | 1000 |
| 4R2QYE | Visual Inspection |  |  |
|  | Sanding | sand paper block | fine |
|  | Polishing/Cleaning | metal surface cleaner |  |
| 4TFF2R | Polishing | Dremel |  |

## TABLE 3

| WebCode | Method | Tool Used | Grit Size |
| :---: | :---: | :---: | :---: |
| 4TG2TD | Acetone |  |  |
|  | Sanding | Sand paper | fine |
|  | Sanding | Sand paper | extra fine |
| 4UVG6U | Sanding | Sand paper | fine |
|  | Polishing | Rotary Tool |  |
| 4WDWEL | Sanding | Sand paper | 400 and 800 |
|  | Polishing |  |  |
| 4XR3XH | Sanding | Emery paper | coarse |
| 672RCF | Polishing | Dremel |  |
| 6DFXMD | Cleaning |  |  |
| 6DYT6E | Sanding, Polishing, and Visual | Sand paper and the etching machine for preperation (Dremel) |  |
| 6EYJLV | Polishing | Dremel |  |
|  | Sanding | Steel wool |  |
|  | Sanding | Sand paper | 320 |
| 6HCJJM | Polishing | Emery paper |  |
| 6KKLGE | Polishing | Rotary Tool |  |
| 6N3UTK | Visual | camera Nikon D90 |  |
|  | Visual | binocular wild M3Z |  |
| 6RNJDX | Visual | Stereoscope |  |
| 6TVVUY | Sanding | Sand paper | 1500-b waterproof sandpaper |
| 6XBJEW | Polishing | Buffing wheel |  |
| 78VZW6 | Sanding | Foredom sanding tool | medium 220 grit |
|  | Polishing | Foredom sanding tool |  |
|  | Visual | Leica Table Microscope |  |
| 7WCH7J | Polishing | Dremel |  |
|  | Cleaning | Water |  |
| 7WPCHD | Cleaning | Dremel | polishing disc |
|  | Sanding | Sand paper | 600 |
| 7YFDPM | Cleaning | Acetone |  |
| 86Y4M2 | Polishing | Dremel |  |
| 88QYAA | None |  |  |

## TABLE 3

| WebCode | Method | Tool Used | Grit Size |
| :--- | :--- | :--- | :--- |
| 8ATXMN | Polishing | Dremel |  |
|  | None | eyes |  |
|  | visual <br> photography <br> sand paper <br> acids <br> photograph <br> sand paper | Polishing | Qtips |

## TABLE 3

| WebCode | Method | Tool Used | Grit Size |
| :---: | :---: | :---: | :---: |
| 9QBD4P | None |  |  |
| 9T3CF3 | I prepared the surface where number has been removed, the preparation includes filing the surface using a filing machine with a fine sanding paper. | Filing machine and sanding paper |  |
| 9XYRDQ | Sanding | Sand paper | 700 |
|  | Polishing | Polishing wheel |  |
| $9 \mathrm{Y76NC}$ | Cleaning | Davis Reagent (Modified) |  |
| A73MD9 | Cleaning | Acetone |  |
| A8ZNN7 | None |  |  |
| AA4JAV | Polishing | Dremel |  |
| ACA28Q | Polishing | Dremel |  |
| ADMNRQ | Visual | eyes |  |
|  | Visual | Stereoscope |  |
|  | Visual | Photographic |  |
|  | Polishing | Dremel |  |
| AEYADL | Sanding | Sand paper | 100 |
|  | Sanding | Sand paper | 360 |
|  | Sanding | Sand paper | 1200 |
|  | Cleaning | Ethanol |  |
| AJDY2N | Sanding | Sand paper | 120 |
|  | Sanding | Sand paper | 400 |
| AJM3G9 | Sanding | Sand paper | 120-fine grit |
| AJRJE7 | Visual | Microscope |  |
| ALYRUH | Polishing | Dremel |  |
| AVGH79 | Sanding | Dremel |  |
| AXBZT4 | None | Stereoscope |  |
| B284XV | Sanding | Sand paper | 220 |
| B4ELVR | Grinding | Dremel |  |
|  | Polishing | Polishing Wheel |  |
| B67GKL | Visual | Microscope |  |
|  | Sanding | Sand paper | 320 |
| BAHHPP | Visual/Micro | Stereoscope |  |
|  | Sanding | Dremel Tool | 180 |

## TABLE 3

| WebCode | Method | Tool Used | Grit Size |
| :---: | :---: | :---: | :---: |
| BHFXRQ | Visual | Stereoscope |  |
|  | Polishing | Dremel |  |
| BWCAZU | Visual | Stereoscope |  |
|  | Visual | Magnet |  |
|  | Cleaning | Isopropanol |  |
|  | Polishing | Dremel | \#425 wheel |
| BWXQFX | Sanding | Sand paper | 200 |
|  | Polishing | Steel wool |  |
| BZCKVU | Visual | Stereoscope |  |
| C4T88E | Polish | Dremel | \#240 wheel |
|  | Polish | Dremel | \#400 wheel |
| C4VPEP | Sanding | Sand paper | 320,600, 1200 |
| C6Q9FR | Sanding | Dremel | 180 |
| CAGUZY | Visual Inspection | Stereo Microscope |  |
|  | Sanding | Foredom K. 2230 | fine sanding (Dremel disc no 4) |
|  | Polishing | Foredom K. 2230 | Fill[sic]-tip polisher |
| CAV438 | None |  |  |
| CCKPJ2 | Polish | Dremel |  |
| CGG7YG | Polishing | Dremel tool |  |
| CNGTNV | Sanding | Dremel |  |
| CYJQW6 | None |  |  |
| CZPHF6 | Visual | Stereoscope |  |
|  | Sanding | Sand paper | fine grit |
|  | Polishing | Dremel |  |
| D8MK7N | Sanding | Sand paper | 100, 400, steel wool |
| DDABPX | Polishing | Dremel | 120,600 |
| DFFTQE | Polishing | Dremel Tool / polis |  |
| DGXMRN | Polishing | Dremel |  |
|  | Sanding | Sand paper | 440 very fine |
| DHNT8E | Visual | Stereoscope |  |
|  | Sanding | Sand paper | fine grit 220 |
| DRF6JV | Sanding | Sand paper | various |
| DTLUKJ | Polishing | Dremel |  |

## TABLE 3

| WebCode | Method | Tool Used | Grit Size |
| :---: | :---: | :---: | :---: |
| DVV2MC | Polishing | Dremel tool |  |
| E2WBVP | Polishing | Dremel/Polishing Disk |  |
| E6Y96W | Polishing | Dremel | 120,600 |
| E77NDN | Visual |  |  |
|  | Polishing | Dremel |  |
| E7QG4Z | Sanding | Sand paper | 100 |
|  | Sanding | Sand paper | 220 |
|  | Sanding | Sand paper | 600 |
|  | Polishing | Steel wool |  |
| E9WUUE | None |  |  |
| EHZ63P | Polishing | Dremel |  |
| EJTXX9 | Polishing | Polishe[sic] |  |
| EJYBJJ | Visual/Microscopic | Stereo microscope |  |
|  | Polishig[sic] | Dremel -w- polishing wheel |  |
| EP43WM | Visual | Stereoscope |  |
|  | Sanding | Sand paper | P600 |
|  | Polishing | Dremel |  |
| EQU347 | Visual | Microscope |  |
|  | Sanding | Sand paper | 400 |
|  | Polishing | Rotary Tool |  |
| ERB8L6 | Polishing | Dremel |  |
| ERPVPH | Visual |  |  |
|  | Sanding |  | 220 grit |
| EUFYDH | Polishing | Rotary Tool |  |
| F33DUT | Visual |  |  |
| F4JCCX | Sanding | Sand paper | 360,400 and 1000 |
| F8VGLC | Visual | Microscope |  |
| F9QZMD | Visual | 5X Eye Loupe |  |
|  | Polishing | Dremel |  |
| F9TPUW | Electro-acid etching method | sand paper electrolytic method device | fine |
| FM9WK2 | None |  |  |
| FR6CHP | Visual | Stereoscope |  |
|  | Grinding | Dremel |  |

## TABLE 3

| WebCode | Method | Tool Used | Grit Size |
| :---: | :---: | :---: | :---: |
| FTX96X | Visual | Microscope | N/A |
|  | Cleaning | Acetone | N/A |
|  | Polishing | Sand paper | P 220 |
|  | Polishing | Sand paper | 1000 |
| FV4TTX | None | N/A | N/A |
| FYLT9D | Cleaning | Acetone |  |
| G7GCPZ | Visual | Stereoscope |  |
| G88KP3 | Clean and polish area of obliteration using a dremel type tool with a sanding disc, and then a polishing disc | Dremel-type tool |  |
| GA4PQQ | Visual | Eyes/Magnifying Glass |  |
|  | Sanding | Sand paper | 220 grit/grain |
| GBXHLA | Visual | Stereoscope |  |
|  | Sanding | Sand paper | 80, 150,400 |
|  | Polishing | Steel wool |  |
|  | Visual | Stereoscope |  |
| GCAAGC | Polishing | Dremel | Jeweler's polishing wheel |
| GF7XVX | Polishing | Emery paper |  |
| GG8LKN | Sanding | Sand paper | 400 |
|  | Polishing | Sand paper | 600 |
| GJAW7Q | Visual | Stereoscope |  |
|  | Sanding | Sand paper | medium |
| GLDFEK | Sanding | Sand paper | 120 |
|  | Sanding | Sand paper | 220 |
| GUMWZY | Cleaning | Acetone |  |
| GVZPED | Polishing | Rotary Tool |  |
| GWW7QB | Visual | Stereoscope |  |
|  | Sanding | Sand paper | fine |
| GWYJMJ | Sanding | Sand paper | fine |
| H24RGT | Magnetic or not | Magnetic Yoke | N/A |
|  | Visual | Desk lamp-magnifier | N/A |
|  | Sanding | Dremel-tool | 240 grit |
|  | Polishing | Dremel-tool | N/A |
| H7CC36 | Polishing | Rotary Tool | FINE GRIT SIZE |
|  | Cleaning |  |  |

## TABLE 3

| WebCode | Method | Tool Used | Grit Size |
| :---: | :---: | :---: | :---: |
| H8BHJY | Sanding | Sand paper | 240 |
| HC7ATG | Sanding | Sand paper | 320 |
|  | Sanding | Sand paper | 600 |
|  | Sanding | Sand paper | 1500 |
|  | Polishing | Simichrome | Simichrome |
| HX6B7K | Sanding | Sand paper | 150, 220, and 400 |
| J46GW8 | Polishing | Dremel | 1200 |
| J4M4UN | Visual | Stereoscope |  |
|  | Polishing | Dremel |  |
| J6ZZX9 | Visual | Stereoscope |  |
|  | Sanding | Sand paper | 150 and 400 |
| JTFAYL | Visual, light source |  |  |
|  | Photograph |  |  |
|  | Hand sanding | Emery paper | 240 grit |
|  | Hand sanding | Crocus cloth | 600 grit |
| JEL2F4 | Filed | File |  |
|  | Emory \& Crocos | Paper | Fine |
|  | Cleaned with Acetone | Qtip |  |
| JEQCKM | Visual | Eyes, maglight |  |
|  | Sanding | Sand paper | 400 |
|  | Sanding | Sand paper | 1200 |
|  | Polishing | Dremel | Rotating polisher |
| JGQNVT | Sanding | Sand paper | 220,400 and 1000 |
| JKEZWJ | Visual |  |  |
|  | Polishing | Dremel |  |
| JKQZQT | Sanding | Sand paper | 360 |
|  | Polishing | Sand paper | 400 |
| JLLHTU | Polishing | Sand paper | 100,220,400 |
| JNECRJ | Visual |  |  |
|  | Polishing | Rotary Tool |  |
| JP7B4W | Dremel Polishing | Dremel | N/A |
| JRXGBF | Sanding | Dremel | Fine |
| JT99GB | Polished | Rotary Tool |  |
| JYWTKQ | Polish | Emory[sic] paper | N/A |
|  | Dremel | polish | \#44 disc |

## TABLE 3

| WebCode | Method | Tool Used | Grit Size |
| :---: | :---: | :---: | :---: |
| K3K6MZ | Sanding | Sand paper | 120 FOLLOWED BY 280 |
| K4KTEC | Visual | Microscope |  |
|  | Polishing | Steel wool |  |
| K4ZEND | Visual | Stereoscope |  |
|  | Polishing | Dremel |  |
|  | Sanding | Abrasive slip stones | medium and fine grit |
| K7PK44 | Polishing | Dremel |  |
|  | Sanding | Sand paper | 400 |
| K9V7WB | Visual/Microscopic | Stereoscope |  |
|  | Polish | Dremel | extra fine wheel |
| KALAKB | Polishing | Sand paper | 400 |
| KCV792 | Sanding | Sand paper | 120 and 80 |
|  | Polishing | Dremel |  |
| KD8UR2 | Sanding | sanding | coarse |
| KRT9VZ | Polishing | Dremel | Fine Texture (Red) |
| KXFWQ7 | Sanding | Sand paper | 320 then 600 |
| KZLGFQ | Note: no visual number was visible. Only machining marks observed. |  |  |
|  | Sanding | Dremil[sic] | 120 |
|  | Sanding \& Polishing | Dremil[sic] | 600 |
|  | Use of alcohol soaked cotton wool to clean bar (before applying electro-acid etching method) | Manual swabbing |  |
| L36WLC | Polishing | Dremel |  |
| L4HKM4 | None |  |  |
| LG4ZQ4 | Polishing | Dremel |  |
| LGNLG9 | Polishing | Dremel |  |
| LGPBMQ | I did little sanding then polished the surface with a soff cloth to be smooth | Dremel tool for sanding | It can be of plus or minus 1 mm in depth |
| LHVXXA | Polishing | Steel wool |  |
| LJQGZB | Polishing | Dremel |  |
| LN3JQD | Polishing | Dremel |  |

## TABLE 3

| WebCode | Method | Tool Used | Grit Size |
| :---: | :---: | :---: | :---: |
| LQD8M7 | Visual | Stereoscope |  |
|  | Polishing | Dremel |  |
| LV4JLA | Polishing | Dremel | medium grit polishing wheel |
| LV92K8 | Visual | Microscope |  |
|  | Visual | Photographic with |  |
| LVJZG3 | Visual | Stereoscope |  |
|  | Polishing | Sand paper |  |
| LVPC3D | Visual |  |  |
|  | Sanding | Sand paper | 320,400 |
| M4TPUZ | Visual | Stereoscope |  |
|  | Polishing | Rotary Tool |  |
| M4Y2FC | Visual | Stereoscope |  |
| M6P8W3 | Sanding | Emery paper | Coarse |
| M73WYE | Sanding | Sand paper | 220 |
|  | Sanding | Sand paper | 320 |
|  | Sanding | Sand paper | 1200 |
|  | Polishing | Steel wool |  |
| M8UVBR | Polishing | Sand paper | Grit size used $\mathrm{N}^{\circ} 220$ and 400 |
|  | Cleaning | Acetone |  |
| MD7UTC | Visual |  |  |
|  | Polishing | Dremel type tool |  |
|  | Visual |  |  |
| MGHQB8 | Polished | Dremel tool | 400 |
| MMNK64 | Polishing | Dremel |  |
| MRK49M | Sanding | Dremer[sic] Tool | 1 mm |
|  | Polishing |  |  |
|  | Visual |  |  |
| MU9FAD | Polishing | Rotary Tool |  |
| MVZCY8 | Polishing | Dremal[sic] |  |
| MWYLQ4 | Polishing | Dremel |  |
| MZYPKC | Visual |  |  |
|  | Microscopic | Stereo Microscope |  |
|  | Sanding | Sand paper | 400 grit |

## TABLE 3

| WebCode | Method | Tool Used | Grit Size |
| :---: | :---: | :---: | :---: |
| N6MWFP | Visual | Microscope | N/A |
|  | Cleaning | Acetone | N/A |
|  | Polishing | Sand paper | P 220 |
|  | Polishing | Sand paper | 1000 |
| N7HFGQ | Sanding | Sand paper | 1000 |
| NF84EE | Polishing | Rotary Tool |  |
| NFJZJJ | Visual | Stereoscope |  |
|  | Sanding | Sand paper | coarse |
|  | Sanding | Sand paper | fine |
|  | Visual | Microscope |  |
| NMGRQR | Visual Inspection | Stereomicroscope |  |
|  | Sanding/Polishing | Sand paper | very fine |
| NQ64RH | Sanding | Dremel | between 120 and 240 |
|  | Sanding | Sand paper | 120 and 240 |
| NRW34V | Sanding | Sand paper | WET/DRY |
| NTUDYY | Polishing | Dremel |  |
| NWTRK3 | Sanding | Dremel | 120 |
|  | Polishing | Dremel | Acessory \#520 |
|  | Sanding | Sand paper | 400 |
| NX7GP2 | Polishing | Dremel |  |
| PAF8BA | Sanding | Sand paper | \#600 Final |
|  | Polishing | Fine loose grit on cloth | Microgrit WCA \# 15 |
| PGC969 | Polishing | Dremel |  |
| PJXXMZ | Sanding | grit | 400 |
|  | Polishing |  |  |
|  | Visual |  |  |
| PL4HDK | light sanding | Dremel tool / hand |  |
|  | Polishing | Dremel tool / hand |  |
| PP8ZPV | electro-acid etching |  |  |
| PWLY6R | Jkm C004 - Examin restoration of erased identification numbers/markings |  |  |
| PZKB4D | Polishing | Dremel |  |
| Q2A4B2 | Sanding | Dremel | NA |
| Q7TYET | Polishing | Polishing wheel |  |

## TABLE 3

| WebCode | Method | Tool Used | Grit Size |
| :---: | :---: | :---: | :---: |
| Q9XLV4 | Polishing | Sand paper | 600 |
| QA9NH3 | Polish |  |  |
| QAT6X7 | visual <br> microscopic polish | stereoscope dremel |  |
| QD9ZD3 | Visual inspection Polishing | N/A <br> Dremel tool |  |
| QLYKGF | Sanding | Sand paper | 220 / very fine |
| QRPVRE | Polishing | Dremel |  |
| QUE4YU | Sanding | Sand paper | fine |
| QY82Q7 | Visual <br> Sanding <br> Polishing |  | $\begin{aligned} & 400 \\ & 600 \end{aligned}$ |
| R6TFRH | Sanding <br> Polishing | Dremel <br> Dremel | Fine |
| RD7CU4 | Sanding <br> Sanding <br> Sanding | Sand paper <br> Sand paper <br> Sand paper | $\begin{aligned} & 100 \\ & 360 \\ & 1200 \end{aligned}$ |
| RDLTPW | Sanding | Sand paper | fine |
| RKPVUW | Polishing | Dremel | Emery wheel on Dremel |
| RL43FE | Sanding | Sand paper | 400 |
| RNQRLD | Polishing | Emery paper | Finished with 1200 grit size. |
| RPG3C7 | Polishing | Dremel |  |
| RR2QUX | The surface was cleaned and burnished | Acetone and sandpaper | 200 |
| RR79TW | Visual <br> Polishing |  |  |
| RZWX6Q | Polishing <br> Visual | Dremel Tool | Fine |
| T6PDDX | Sanding <br> Polishing | Sand paper <br> Sand paper | $\begin{aligned} & 150 \\ & 400 \end{aligned}$ |
| TC8L4F | Sanding | Dremel | fine |

## TABLE 3

| WebCode | Method | Tool Used | Grit Size |
| :---: | :---: | :---: | :---: |
| TKZ48P | Visual | Eyes |  |
|  | Sanding | Sand paper | 32 mm diameter |
|  | Polishing | Polishing paper | 32 mm diameter |
|  | Acids | Nitric acid, Distilled water, Hydrochloric Acid, Copper (II) ammonium Chloride |  |
| TQ7MH2 | Polishing | Dremel |  |
| TXN2G2 | Polishing | Dremel |  |
| TXQQNJ | Polishing | Dremel | 120/600 |
| U2YFFQ | Sanding | Sanding paper | 800 |
| U8BEXC | None |  |  |
| UDWAYF | I used a little bit of a sander then polished the surface with a soft cloth to smooth the surface | (Dremel) sander for polishing | it can be of plus/minus 1 mm in depth. |
| UJ34VW | Polishing | Dremel |  |
| UP7YQP | Visual method | Stereo microscope |  |
|  | Sanding \& Polishing | Sand paper \& Polisher | fine |
| UVWA2N | None |  |  |
| UYWKWN | Polishing | Dremel |  |
| UZNR37 | Polishing | Dremel | N/A |
| V62RZX | Polish | Dremel Tool |  |
| VE976K | Visual | Microscope | - |
| VJZQV8 | Polishing | Dremel |  |
| vM3CZ8 | Sanding | Sand paper | 150,400 |
|  | Polishing | Steel wool |  |
| VP6ABZ | Polishing | Dremel |  |
| VRA33D | Sanding | Sand paper | 360 \& 307 |
| VYBGNE | Lightly polished the obliterated surface area | Dremel Polisher | 1 mm |
| VZ68UT | Sanding | Sand paper | P400 |
| W2F8QM | Visual | Stereoscope |  |
| W4LRF9 | Polishing | Dremel |  |
| W8L3AM | Polishing | Dremel | fine cratex wheel |

## TABLE 3

| WebCode | Method | Tool Used | Grit Size |
| :---: | :---: | :---: | :---: |
| WBJJMG | Polishing | Rotary Tool |  |
|  | Cleaning | Acetone |  |
| WCARLK | Sanding | Dremel | 438 band |
|  | Polishing | Dremel |  |
| WD8VN9 | Sanding | Sand paper | P150 \& 220 |
|  | Polishing | Sand paper | 320 \& 600 |
|  | Polishing | Dremal[sic] |  |
| WFA3HK | Visual | Stereoscope |  |
|  | Polishing | Rotary Tool |  |
|  | None |  |  |
| WGAUYZ | Visual | N/A |  |
| WL22EN | None |  |  |
| WL79U3 | Polishing | Dremel |  |
| WQWTP7 | Polishing | Dremel |  |
| WRRBR8 | Polishing | Dremel |  |
| WTK8DF | Lightly polished the obliterated area | Dremel polisher | 1 mm |
| WTMY7W | Visual | Stereoscope |  |
|  | Cleaning | Acetone |  |
|  | None |  |  |
| WXJHC2 | Visual | Stereoscope |  |
|  | Polishing | Dremel |  |
| XA4J3R | Visual | Stereoscope |  |
|  | Sanding | Sand paper | 150 \& 400 |
| XNM3Z3 | Polishing | Dremel |  |
| XQEYNW | Polish | Dremel tool |  |
| Y8DM7L | Sanding and Polishing with SiC Paper |  | 400 grit and 120 grit |
| Y96TC4 | None |  |  |
| YBWP2X | Visual | Stereoscope |  |
|  | Sanding | Sand paper | 320,400,1200 |
| YKDLTH | Visual | Stereoscope |  |
|  | Polishing | Dremel |  |
| YKNYU2 | Grinding | Dremel | N/A |
| YMPC8T | Polishing | Dremel |  |

## TABLE 3

| WebCode | Method | Tool Used | Grit Size |  |
| :---: | :---: | :---: | :---: | :---: |
| YXBFER | Polishing | Dremel |  |  |
| Z9JKYT | Polishing | Sand paper | 150 |  |
|  | Sanding | Sand paper | 150 |  |
| ZAXZG2 | Sanding | Dremel | 240 |  |
| ZDHU9V | Visual | Stereoscope |  |  |
| ZE9ZPL | Sanding | Sand paper | medium |  |
| ZJMXW9 | Polishing | Dremel |  |  |
| ZL47ZE | Polishing | Emery paper |  |  |
| ZX3V4U | Sanding | Sand paper | 80-180 |  |
|  | Polishing | Dremel |  |  |
| ZYVNYD | Visual | Microscope | N/A |  |
|  | Cleaning | Acetone | N/A |  |
|  | Polishing | Sand paper | P220 |  |
|  | Polishing | Sand paper | 1,000 |  |
| ZZWCN4 | Polishing | Emery paper | 1200 |  |
|  | None | M7V1 Posi test |  |  |
|  | Visual | M2V1 Cast |  |  |
| Response Summary |  |  |  | Participants: 302 |
| Sample Preparation |  |  |  |  |

Visual Method: 90
Sanding Method: 114
Polishing Method: 174
None: 22

Note: The totals are not equivalent to the total number of participants because some participants used more than one sample preparation method.

## Recovery Methods

## (listed in order of use)

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| 26JQKN | Phosphoric/Nitric Acid then 25\% Nitric with some rubbing of surface. | 3 minutes / 1 minute |
| 27EALP | Acid Etch Method | Nitric acid 25\%, 5 minutes |
| 2ACVB8 | Magnaflux |  |
|  | Acidic Ferric Chloride | approximately 15 minutes |
|  | 10\% Sodium Hydroxide | approximately 2 minutes |
|  | Acid Etch Method | approximately 5 minutes (20\% Nitric Acid) |
| 2B8DD9 | Ecid[sic] etching | +/-15 min |
| 2BNLMW | Zinc Reagent \#1 (H3PO4/HNO3) | 1-2 minutes |
|  | Zinc Reagent \#2 (dilute HNO3) | 30-45 seconds |
| 2CJ4NX | Ferric Chloride | Spent a total of $21 / 2$ days on bar stock |
|  | Acidic Ferric Chloride |  |
|  | Phosphoric/Nitric Acid |  |
|  | Nitric Acid |  |
|  | Sodium Hydroxide |  |
|  | Fry's Reagent |  |
| 2NTQPH | Atlanta 2 Step Phosphoric/Nitric Acid | 2 minutes total |
| 2P7FRV | Zink[sic] Alloy Acid Etch (Phosphoric Acid \& Nitric Acid) | Pool \& wipe $+/-45$ minutes total |
| 2QJRWR | Acid Etch Method | Alternate application between Phosphoric and Nitric Acids until characters developed, approximately 8-10 minutes. |
| 2QXCGP | Acidic Ferric Chloride <br> Ferric Chloride <br> touch of Turner's \& NaOH | $\sim 5$ min total <br> (in multiple applications) |
| 2QZ9ME | Acid Etch Method | 10-15 seconds |
| 2WKBKH | 25\% Nitric Acid | $\sim 2$ minutes |
| 2XW2XT | Fry's Reagent | 10/sec per use |
| 2ZQHH3 | Acid Etch Method | Lab method - M8 V1 - Approx 50 60 mins |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| $32 \mathrm{BX2A}$ | Acidic Ferric Chloride | 20 minutes |
|  | Ferric Chloride |  |
|  | Sodium hydroxide |  |
|  | Nitric acid | 20 minutes |
| 33R6BX | Acidic Ferric Chloride | ~ 100 swabs |
| 34923X | Acidic Ferric Chloride | 5 min |
|  | Turner's Reagent | 5 min |
|  | Fry's Reagent | 5 min |
| 3836 JH | Acidic Ferric Chloride | 20-30 SEC PER APPLICATION |
| 3B64WW | Chemical Etching | material was swabbed etchant not left standing on material |
|  | Ferric Chloride \& 25\% Nitric Acid | $<5$ minutes |
| $3 B Z Q B L$ | Sodium Hydroxide 10\% | Few seconds per swab. |
|  | Ferric Chloride | Few seconds per swab. |
|  | Acidic Ferric Chloride | Few seconds per swab. |
|  | Nitric Acid 25\% | Few seconds per swab. |
| 3D6C27 | Acid Etch Method | Approximately 30 seconds (20\% Nitric Acid) |
|  | Acidic Ferric Chloride | Approximately 30 seconds |
|  | Acid Etch Method | Approximately 30 seconds (20\% Nitric Acid) |
| $3 M U Z X U$ | Acid etching - Zinc Alloy Solutions (1 \& 2) | continuous swabbing - less than one minute |
|  | Acid etching - Ferric Chloride | continuous swabbing - approx one minute |
|  | Acid etching - Zinc Alloy Solutions (1 \& 2) | continuous swabbing - approx five minutes |
| $3 N P H Z V$ | Acid Etch Method | two minutes |
| 3PK7EK | Fry's Reagent | 30 MINUTES |
| 3 QFPFL | Acidic Ferric Chloride | 15 MINUTES |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| 3RTGBN | Acidic Ferric Chloride | multiple swipes over 10 minutes |
|  | 25\% Nitric Acid | one swipe to highlight |
|  | Davis Reagent | multiple swipes over 3 minutes |
|  | 25\% Nitric Acid | one swipe to highlight |
|  | Turner's Reagent | multiple swipes over 20 minutes |
|  | polishing | polish to clean |
|  | Fry's Reagent | multiple swipes over 2 minutes |
|  | 25\% Nitric Acid | one swipe to highlight |
|  | Fry's Reagent | multiple swipes over 2 minutes |
| 3UD7U2 | Acid Etch Method | swabbed |
| 3ZJYPW | Acid Etch Method | left on surface for intervals of 20-50 seconds before wiped off then swiped over area for $5-10$ seconds |
|  | Acidic Ferric Chloride | swiped across the surface for 5-10 seconds followed by 10 second application of $25 \%$ nitric acid |
| 438NV | Acid Etch Method | $\sim 1 \mathrm{sec}$ or less |
| 477ZQV | acidic ferric chloride acid etch | $\sim 2$ mins. |
|  | 25\% nitric acid etch | $\sim 1 \mathrm{~min}$. |
| 482HTW | Acidic Ferric Chloride | 30 minutes |
| 4GK94M | Ferric Chloride | a few seconds |
|  | 25\% Nitric Acid | a few seconds |
| 4LE34A | Acid Etch Method | Total 44 minutes |
|  | Ferric Chloride | 15 minutes |
|  | Acidic Ferric Chloride | 15 minutes |
|  | Ferric Chloride | 12 minutes |
|  | Acidic Ferric Chloride | 2 minutes |
| 4R2QYE | Ferric Chloride/Nitric Acid | continuous for approx 30 minutes |
| 4TFF2R | Acidic Ferric Chloride | Less than one minute |
|  | Ferric Chloride | Less than one minute |
|  | Fry's Reagent | Less than one minute |
| 4TG2TD | Ferric Chloride (Acid Etch) | seconds (few) |
|  | Nitric Acid (acid etch) | seconds (few) |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| 4UVG6U | sand paper |  |
|  | Acidic Ferric Chloride | 30 seconds to 1 minute |
|  | deionized water rinse |  |
|  | polishing |  |
|  | Acidic Ferric Chloride | 30 seconds to 1 minute |
|  | deionized water rinse |  |
|  | sand paper |  |
|  | Acidic Ferric Chloride | 30 seconds to 1 minute |
|  | deionized water rinse |  |
|  | polishing |  |
|  | Acidic Ferric Chloride | 30 seconds to 1 minute |
|  | deionized water rinse |  |
|  | 10\% HCl | 30 seconds to 1 minute |
|  | polishing |  |
|  | 10\% HCl | 30 seconds to 1 minute |
|  | deionized water rinse |  |
|  | Acidic Ferric Chloride | 30 seconds to 1 minute |
|  | deionized water rinse |  |
| 4WDWEL | Fry's Reagent | 5 mn |
| 4XR3XH | Acidic Ferric Chloride | 2 min . |
| 672RCF | Phosphoric/Nitric acid solution | 10 minutes |
|  | Dilute Nitric acid solution | 10 minutes |
| 6DFXMD | Other |  |
| 6DYT6E | Acid Etch | five seconds |
| 6EYJLV | Acid Etch Method | Ferric Chloride - 1 minute |
|  | Acidic Ferric Chloride | approximately 30 seconds |
|  | Acid Etch Method | 10\% Nitric Acid - 1 minute |
|  | Acid Etch Method | Ferric Chloride - 1 minute 30 seconds |
|  | Acid Etch Method | 10\% Nitric Acid - 2 minutes |
| 6HCJJM | Acid Etch Method | 30 min |
| 6KKLGE | Electro-acid |  |
| 6N3UTK | Acid Etch Method | $3 \times 45$ " |
| 6RNJDX | Electrolytic - 10\% sodium hydroxide | Characters were brought up in less than 2 minutes. |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| 6TVVUY | Acidic Ferric Chloride | One minute |
|  | 25\% Nitric Acid swab | 15 seconds |
|  | Acidic Ferric Chloride | One minute |
|  | 25\% Nitric Acid swab | 15 seconds |
|  | Acidic Ferric Chloride | One minute |
|  | 25\% Nitric Acid swab | 15 seconds |
|  | Rinse with distilled water | Rub dry |
| 6XBJEW | Magnetic Particle Inspection (MPI) |  |
|  | Fry's Reagent | 5 minutes |
|  | Hydrochloric acid solution | 5 minutes |
|  | Nitric acid solution | 5 minutes |
| 78VZW6 | Acid Etch | Until number became visible |
| 7WCH7J | Phosphoric/Nitric Acid | 5-8 minutes |
|  | 25\% Nitric Acid | 30 seconds - 1 minute |
| 7WPCHD | Fry's Reagent |  |
|  | Acidic Ferric Chloride |  |
| 7YFDPM | Zinc Alloy Solutions (2-step) | 5-6 rounds of solutions \#1 and \#2, approximately 3 minutes per round. |
| 86Y4M2 | Acid Etch Method | 4 swabs (Acidified Ferric Chloride) |
|  | Acid Etch Method | 10 swabs (Nitric Acid) |
| 88QYAA | Clean cotton wool swab using 10\% sodium hydroxide solution. |  |
| 8ATXMN | Fry's Reagent | less than 5 seconds |
|  | Acid Etch Method | 30 second intervals |
|  | Acidic Ferric Chloride | 30 second intervals |
| 8DEM6F | sand paper to smooth p220 |  |
|  | then used Nitric Acid | 2 min |
|  | an acidic ferric reagent | 2 min |
|  | Then used together |  |
| 8GC4GA | $25 \%$ Ferric chloride solution applied to surface for a few seconds. No characters appeared. |  |
|  | 25\% Nitric Acid was applied to the surface. Characters appeared immediately. | three seconds |
| 8XYGXM | Acid Etch Method | Ferric chloride |
|  | Fry's Reagent | Very dilute Fry's |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| 92WHBP | Acid Etch Method | three applications of about one minute each |
| 94H7TG | HCL 50\%-DISTILLED WATER 50\% |  |
| 96T7RW | Acid Etch Method | 1-2 seconds at a time |
| 97A3JV | Acidic Ferric Chloride | 3 min |
|  | Cobalt Chloride | 3 min |
|  | Nitric Acid | 5 min |
| 97NPTX | 25\% Nitric Acid | 3-5 min |
|  | Ferric Chloride | $3-5 \mathrm{~min}$ |
|  | 25\% Nitric Acid | $2-4 \mathrm{~min}$ |
|  | Ferric Chloride | $3-5 \mathrm{~min}$ |
| 99XRPU | Acidic Ferric Chloride | 2-3 hours |
|  | 25\% nitric acid for contrast | a few seconds |
| 9E3B6M | Acid Etch Method | Hf - less than 1 minute |
| 9EKJTG | Restor-A-Gel - Aluminum (Mercuric Chloride) | Approximately 1 hour 45 minutes |
| 9FY4LR | Ferric Chloride | 45 min |
|  | Nitric Acid | 15 min |
| 9K4Y98 | Acid Etch Method | 2 min . |
| 9KN9QN | miproacies[sic] | 15 min . |
| 9L3XU2 | Ferric Chloride |  |
|  | Acidic Ferric Chloride | 2-4 sec |
|  | Nitric Acid 25\% | swabbed |
|  | NaOH (25\%) |  |
| 9MVQPK | Acid Etch Method | maximum- around 5 minutes |
| 9N8HJM | Acid Etch Method | 3-5 min |
|  | Acid Etch Method | $1 \text { min }$ |
| 9Q3Y6W | Acid Etch Method | HCL applied for 45 minutes |
| 9QBD4P | Acid Etch Method | swabbed on area, not ponded (pooled) |
| 9T3CF3 | Electro-acid etching process | 25 seconds |
| 9XYRDQ | MagnaFlux |  |
|  | $10 \% \mathrm{NaOH}$ |  |
|  | Ferric Chloride |  |
|  | Acidic Ferric Chloride | $<1$ minute |

TABLE 4

| WebCode | Method | Time |
| :--- | :--- | :--- |
| 9Y76NC | Electrochemical treatment using Davis <br> Reagent (Modified) were used about <br> one (1) minute | One (1) minute |
|  | Etching reagent: Cupric Ammonium <br> Chloride solution |  |
| A73MD9 | 10\% Sodium Hydroxide Solution | 5 minutes |
|  | Acid Etch Method | 45 minutes |
| A8ZNN7 | Acid Etch Method | 5 min. $10 \%$ NaOH electrolytic |
|  | Acid Etch Method | 1 min. $25 \%$ Nitric Acid |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| BHFXRQ | Acidic Ferric Chloride | 1 MIN |
|  | 25\% Nitric Acid | 1 MIN |
|  | 50\% Nitric Acid | 1 MIN |
| BWCAZU | Ferric Chloride | $<1$ min |
|  | Acidic Ferric Chloride | $\sim 5 \mathrm{~min}$ @ 30 second intervals |
|  | Nitric Acid (25\%) | <1 minute total |
| BWXQFX | Fry's Reagent | 30 sec |
|  | Acid Etch Method | 10 min |
| BZCKVU | Acidic Ferric Chloride | 2-3 minutes |
|  | 25\% Nitric Acid | 2-3 minutes |
| C4T88E | Chemical Restoration - Chromic Acid | 2 minute increments - 10 min total |
| C4VPEP | Acid Etch Method | approx $5-10$ seconds |
| C6Q9FR | Visual |  |
|  | Sanding |  |
|  | Ferric Chloride | $\sim 15$ seconds |
|  | Acidic Ferric Chloride | $\sim 30$ seconds |
|  | Nitric Acid | $\sim 15$ seconds |
| CAGUZY | Acid Etch | +/-5 minutes |
| CAV438 | 10\% Nitrate Acid | approximately 5 min |
| CCKPJ2 | Ferric Chloride Reagent | 5 min |
|  | Nitric Acid 25\% | 10 min |
|  | Sodium Hydroxide | 5 min |
| CGG7YG | Chemical etching | approx 15 minutes |
| CNGTNV | 25\% Nitric Acid | 2 mins |
| CYJQW6 | Etching using $10 \% \mathrm{NaOH}$ |  |
| CZPHF6 | Acid Etch Method | 25\% HNO3 ~ 5 mins |
|  | Acid Etch Method | 50\% HNO3 ~ 10 mins |
|  | Acid Etch Method | $\mathrm{FeCl} 3-\mathrm{H} 2 \mathrm{O} \sim 3 \mathrm{mins}$ |
|  | Acid Etch Method | $10 \% \mathrm{NaOH} \sim 14 \mathrm{mins}$ |
| D8MK7N | MagnaFlux |  |
|  | Acid Etch Method | 10-30 seconds at a time for approx 7 10 minutes. |
| DDABPX | Nitric Acid | +/-2 minutes |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| DFFTQE | Acid Etch $85 \% \mathrm{H} 3 \mathrm{PO} 4 / \mathrm{HNO} 3$ | 10-15 seconds |
|  | Acid Etch 20\% HNO3 | 10-15 seconds |
|  | Acid Etch 10\% HNO3 | 10-15 seconds |
| DGXMRN | Acid Etch Method | 10-15 Minutes |
| DHNT8E | Acidic Ferric Chloride | 1 hour |
|  | Other |  |
| DRF6JV | Chemical Etch |  |
|  | Ferric Chloride | 2 mins |
|  | Acidic Ferric | 1 min |
|  | Nitric Acid | 30 seconds |
|  | Sodium Hydroxide | 1 min |
| DTLUKJ | Acidic Ferric Chloride | 10 seconds |
| DVV2MC | Chemical Etching - Ferric Chloride | Several seconds to several minutes |
|  | Chemical Etching - Acidic Ferric Chloride | Several seconds to several minutes |
|  | Chemical Etching - Ferric Chloride | Several seconds to several minutes |
|  | Chemical Etching-25\% Nitric Acid | Several seconds to several minutes |
| E2WBVP | Acidic Ferric Acid | 5 min |
|  | Cobalt Chloride | 30 sec |
|  | Acidic Ferric Acid | 20 min |
|  | Acidic Ferric Acid (Diluted) | 5 min |
| E6Y96W | Nitric Acid | +/- 2 min |
| E77NDN | Fry's Reagent | 1 min |
|  | Nitric Acid | 2-3 min |
|  | Phosphoric Acid | 1-2 min |
| E7QG4Z | Acidic Ferric Chloride | 25 minutes |
|  | Acid Etch Method | 5 minutes |
| E9WUUE | WAZAU (CuCl2, Na CL) | 3 minutes |
| EHZ63P | MagnaFlux |  |
|  | Ferric Chloride | Not Recorded. Extimated[sic] to be 1 to 3 minute intervals. |
|  | Nitric Acid | Not Recorded. Extimated[sic] to be 1 to 3 minute intervals. |
| EJTXX9 | Fry's Reagent | Ten minutes |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| EJYBJJ | Phosphoric acid/nitric acid solution | $\sim 10$ seconds |
|  | 25\% nitric acid solution | $\sim 5$ seconds |
|  | Phosphoric acid/nitric acid | $\sim 30$ seconds |
|  | 25\% nitric acid | $\sim 5$ seconds |
|  | Phosphoric acid/nitric acid | $\sim 10$ seconds |
|  | 25\% nitric acid | $\sim 3$ seconds |
| EP43WM | 25\% Nitric Acid | 2-3 min |
| EQU347 | Ferric Chloride | 10-15 seconds |
| ERB8L6 | Acid Etch Method | 2-5 minutes |
|  | Griffin Reagent | 20 seconds |
|  | 10\% Sodium Hydroxide | 5 minutes |
|  | Griffin Reagent | 20 seconds |
|  | Fry's Reagent | 10 seconds |
|  | Griffin Reagent | 20 seconds |
|  | Acid Etch Method | 2-5 minutes |
|  | Acid Etch Method | 5 minutes |
|  | Electrolytic (10\% Sodium Hydroxide) | 15 minutes |
|  | Acid Etch Method | 5 minutes |
| ERPVPH | Ferric Chloride | alternated between the two |
|  | 50\% Nitric Acid | about ten seconds each |
| EUFYDH | Acidic Ferric Chloride water | 30 seconds |
|  | Acidic Ferric Chloride wiped with tissue | 30 seconds |
|  | Acidic Ferric Chloride wiped with tissue | 30 seconds |
|  | Acidic Ferric Chloride wiped with tissue | 30 seconds |
| F33DUT | Acid Etch Method | 5 seconds (20\% Nitric Acid) |
|  | Acid Etch Method | 3 seconds (Acidic Ferric Chloride) |
| F4JCCX | Ferric chloride $27 \% \mathrm{~m} / \mathrm{v}$ | Not applied |
|  | Acidic Ferric Chloride | Less than a minute |
|  | Sodium hydroxide $13.5 \% \mathrm{~m} / \mathrm{v}$ | Not applied |
|  | Nitric acid 25\% | Less than a minute |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| F8VGLC | Ferric Chloride | Swabbed ( $\sim 15$ mins) |
|  | Acidic Ferric Chloride | Swabbed ( $\sim 15$ mins) |
|  | Nitric Acid (25\%) | Swabbed ( $\sim 5$ mins) |
|  | Polishing (done in between swabbing) | One time ( $\sim 10 \mathrm{sec}$ ) |
| F9QZMD | Acid Etch Method |  |
|  | Pot Metal Sol. 1 | 10 Seconds |
|  | Aqua Regia | 20 Seconds |
|  | 25 \% Nitric Acid | 20 Seconds |
| F9TPUW | Electro-acid etching method | +/-10 min |
| FM9WK2 | 25\% Nitric Acid | 3-5 Minutes |
| FR6CHP | Ferric Chloride | 5 Minutes |
|  | Acidic Ferric Chloride | 5 Minutes |
|  | 25\% Nitric Acid | 2 Minutes |
| FTX96X | Acid Etch Method | Total: 23 minutes |
|  | Ferric Chloride | 16 minutes |
|  | Acidic Ferric Chloride | 7 minutes |
| FV4TTX | Ferric Chloride | $\sim 5$ minutes |
|  | Acidic Ferric Chloride | $\sim 5$ minutes |
|  | Phosphoric / Nitric Acid | $\sim 20$ minutes |
|  | 25\% Nitric Acid | $\sim 2$ minutes |
| FYLT9D | Electrolytic etching using $10 \%$ sodium hydroxide. |  |
| G7GCPZ | 10\% aqueous Sodium Hydroxide |  |
| G88KP3 | After polishing/sanding, clean surface with acetone/methanol then apply acid etch (chemical Green Mamba) with cotton swabs and observe for appearance of numbers and wipe acid off with cotton swab (dipped in methanol), <br> [continued] continue process if/when some digits aren't visable. | In total, about 20 minutes or more |
| GA4PQQ | Ferric Chloride | several seconds at a time |
|  | Acidic Ferric Chloride | several seconds at a time |
|  | 25\% Nitric Acid | several seconds at a time |
|  | 10\% Sodium Hydroxide | several seconds at a time |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| GBXHLA | Acid Etch Method | Ferric Chloride 10 min |
|  | Acidic Ferric Chloride | 10 min |
|  | Acid Etch Method | Alternation Ferric Chloride \& 25\% Nitric Acid 5 min each for 100 min |
|  | Acid Etch Method | $25 \%$ Nitric Acid for 5 min with no cotton swab |
| GCAAGC | 25\% Nitric Acid and Phosphoric Acid (2 step) | Constantly moving applicator-5 mins |
| GF7XVX | Other | less of a minute |
| GG8LKN | Fry's Reagent | 5 minutes in total |
| GJAW7Q | Fry's Reagent | a few seconds |
|  | 25\% Nitric Acid | a few seconds |
| GLDFEK | Sodium Hydroxide | $\sim 1$ hour (total) |
|  | Ferric Chloride | 30-40 sec each app. |
|  | Fry's Reagent | 15-30 sec each aplication[sic] |
|  | Nitric Acid 25\% | 15-20 sec. each application |
|  | Acidic Ferric Chloride | 15-20 sec. each application |
| GUMWZY | Fry's Reagent |  |
| GVZPED | Acidic Ferric Chloride |  |
|  | 25\% Nitric Acid |  |
|  | Acidic Ferric Chloride |  |
|  | Phosphoric/Nitric Acid |  |
|  | Acidic Ferric Chloride |  |
|  | 25\% Nitric Acid |  |
|  | Acidic Ferric Chloride |  |
| GWW7QB | Acid Etch Method | Ferric Chloride |
|  | Acid Etch Method | 10\% Sodium Hydroxide |
|  | Acid Etch Method | Ferric Chloride |
|  | Acid Etch Method | 25\% Nitric acid |
| GWYJMJ | Acidic Ferric Chloride | $\sim 30$ sec increments, $\sim 20 x^{\prime}$ s |
|  | Fry's Reagent | $\sim 10$ sec increments, $\sim 5 x^{\prime}$ s |
|  | Acidic Ferric Chloride/ NaOH | $\sim 30 \mathrm{sec} / \sim 10 \mathrm{sec}, \sim 5 \mathrm{x}^{\prime} \mathrm{s}$ |
|  | MagnaFlux |  |
|  | FeCl3/25\% HNO3 | $\sim 30 \mathrm{sec} / \sim 30 \mathrm{sec}, \sim 20 \mathrm{x}$ 's |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| H24RGT | Acid Etch - Flicks | 3 seconds |
|  | Acid Etch - Green Mamba | 2 seconds |
|  | Electro-Acid Etch - Green Mamba | 2 seconds |
| H7CC36 | PALMERTON REAGENT | 35 MINUTES |
| H8BHJY | Acidic Ferric Chloride | 30-60 second for 10 times, alternate wash with distilled water. |
| HC7ATG | Acid Etch Method | less than a minute |
| HX6B7K | Acid Etch Method | 5 minutes |
| J46GW8 | Acid Etch Method | 2 minutes |
| J4M4UN | Acidic Ferric Chloride | 5-10 seconds for each application |
| J6ZZX9 | 10\% Sodium Hydroxide |  |
|  | Ferric Chloride |  |
|  | Acidic Ferric Chloride | 30 seconds |
|  | Ferric Chloride |  |
|  | 25\% Nitric Acid (pooling) | 20 minutes |
| J7FAYL | Acid Etch |  |
|  | Swab using 10\% Nitric Acid solution | 3 minutes |
| JEL2F4 | Acid Etch | Total 20 minutes |
|  | Acetone between Acid Etch |  |
| JEQCKM | $5 \% \mathrm{HCl}$ in ethyl alcohol | 15 min nil react. |
|  | $5 \% \mathrm{HCl}$ aqueous | Metal lighten, 30 min |
|  | $50 \% \mathrm{HCl} / 50 \% \mathrm{HNO} 3$ alternately used | 60 min |
| JGQNVT | Acidic Ferric Chloride | Less than a minute |
|  | Ferric Chloride | Less than a minute |
|  | Sodium Hydroxide $13.5 \mathrm{~m} / \mathrm{v}$ | Less than a minute |
|  | Nitric Acid 25 \% v/v | Less than a minute |
| JKEZWJ | Ferric Chloride | 1-2 mins. |
|  | Acidic Ferric Chloride | 1-2 mins. |
|  | Nitric Acid | 2-3 mins. |
| JKQZQT | Acidic Ferric Chloride | Less than a minute |
|  | Ferric chloride | Not apply |
|  | Hydrochloric acid 25\% | Less than a minute |
| JLLHTU | Acid Etch Method | HCl 10 minutes |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| JNECRJ | Acidic Ferric Chloride | 1-2 minutes |
|  | Nitric Acid | 4 minutes |
| JP7B4W | Ferric Chloride | $\begin{aligned} & (2: 30) 1 \mathrm{~min}, 20 \mathrm{sec}, 20 \mathrm{sec}, 20 \mathrm{sec}, 30 \\ & \mathrm{sec} \end{aligned}$ |
|  | Acidic Ferric Chloride | $(16 \mathrm{~min}) 1 \mathrm{~min}, 2 \mathrm{~min}, 1 \mathrm{~min}, 2 \mathrm{~min}, 3$ $\mathrm{min}, 1 \mathrm{~min}, 3 \mathrm{~min}, 1 \mathrm{~min}, 2 \mathrm{~min}$ |
|  | 10\% Nitric Acid | (1:40) $20 \mathrm{sec}, 20 \mathrm{sec}, 30 \mathrm{sec}, 30 \mathrm{sec}$ |
| JRXGBF | Ferric Chloride Solution | 45 minutes |
|  | Acidic Ferric Chloride | 30 minutes |
|  | 10\% Sodium Hydroxide | 5 minutes |
|  | 10\% Nitric Acid | 5 minutes |
| JT99GB | Aluminum Solution 017-16-01 | 1 minute |
|  | Phosphoric Nitric 008-16-01 | 1 minute |
| JYWTKQ | Ferric Chloride 100\% | 2 min. no response |
|  | Nitric Acid 25\% | 5 min. |
|  | Sodium Hydroxide 10\% | 2 min . for contrast |
| K3K6MZ | Acid Etch Method | 2-4 MINUTES |
| K4KTEC | Ferric Chloride | 2 applications at 2 minutes. |
| K4ZEND | 25\% Nitric Acid | Instantaneous reaction and faded after time |
| K7PK44 | Acidic Ferric Chloride | 15 MINUTES |
|  | CHEMICAL HIDROCHLORIC ACID 25\% | 10 MINUTES |
| K9V7WB | Nitric Acid 25\% | 2 minutes (performed twice) |
|  | Sulfuric Acid 30\% | 30-45 seconds |
| KALAKB | Fry's Reagent | 5 minutes |
| KCV792 | Acid Etch Method | Acid was wiped over with a cotton bud for approximately 5 minutes. |
| KD8UR2 | Acidic Ferric Chloride | 2 mins |
| KRT9VZ | Phosphoric Acid | 8 min |
|  | Nitric Acid | 2 min |
| KXFWQ7 | Acid Etch Method | 30 seconds |
| KZLGFQ | Electro-acid etching (with Green Mamba acid solution \& electric current.) | $+/-2$ minutes altogether. Acid applied at 4 intervals of 30 seconds each, with deactivation (via alcohol swabs) in between each acid application. |
| L36WLC | FeCl 3 | 10 Min |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| L4HKM4 | Sodium Hydroxyde[sic] 10\% | 10 sec |
|  | Acid Etch Method | 1 min |
| LG4ZQ4 | Acidic Ferric Chloride | 30 seconds |
|  | Phosphoric Nitric Acid | 2 minues |
|  | 25\% Nitric Acid | 30 seconds |
| LGNLG9 | Acid Etch Method |  |
|  | PHOSPHORIC ACID/NITRIC ACID | 15 MINUTES |
|  | NITRIC ACID | 10 MINUTES |
| LGPBMQ | Electro-acid etching process was used to recover obliterated serial numbers | It took about + /- 10 mins |
| LHVXXA | Acidic Ferric Chloride |  |
|  | Fry's Reagent |  |
| LJQGZB | $10 \% \mathrm{NaOH}$ | 10 minutes with cotton swab |
|  | Acidic Ferric Chloride | 10 minutes with cotton swab |
| LN3JQD | Ferric Chloride | ~10-20 sec while wiping |
|  | Nitric Acid | $\sim 2-5$ sec while wiping |
|  | Acidic Ferric Chloride | $\sim 5-10$ seconds once |
| LQD8M7 | Ferric Chloride |  |
|  | Acidic Ferric Chloride |  |
|  | 25\% Nitric acid |  |
| LV4JLA | 25\% nitric acid | 1-2 minutes |
| LV92K8 | Acidic Ferric Chloride | 45 minutes |
|  | 25\% Nitric Acid | 45 minutes |
|  | 10\% Sodium Hydroxide | 45 minutes |
| LVJZG3 | Chemical Method with Zinc alloy etching | 1 hour |
| LVPC3D | Acid Etch Method | Knowles Process - 85\% phosphoric acid, concentrated nitric acid |
|  | Acid Etch Method | 25\% Nitric Acid |
| M4TPUZ | Hidrochloric acid | Approximately two minutes |
| M4Y2FC | Acid Etch Method | few seconds at the time, several times, alltogether[sic] 10-15 min |
| M6P8W3 | Acidic Ferric Chloride | $5-10$ seconds per application |
| M73WYE | Ferric Chloride | $\sim 10 \mathrm{~min}$ |
|  | Acidic Ferric Chloride | $\sim 3 \mathrm{~min}$ |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| M8UVBR | Fry's Reagent | About 5 minutes |
| MD7UTC | Magnaflux |  |
|  | Acidic Ferric Chloride | $\sim 30 \mathrm{sec}$ |
|  | Acidic Ferric Chloride | $\sim 30 \mathrm{sec}$ |
|  | Dremel type tool | polishing |
|  | Acidic Ferric Chloride | $\sim 30 \mathrm{sec}$ |
|  | Acidic Ferric Chloride | $\sim 30 \mathrm{sec}$ |
|  | Dremel type tool | polishing |
|  | 25\% Nitric Acid | $\sim 10 \mathrm{sec}$ |
|  | Dremel type tool | polishing |
|  | Acidic Ferric Chloride | $\sim 30 \mathrm{sec}$ |
|  | Acidic Ferric Chloride | $\sim 30 \mathrm{sec}$ |
|  | Dremel type tool | polishing |
|  | NaOH | $\sim 10 \mathrm{sec}$ |
|  | Dremel type tool | polishing |
|  | Acidic Ferric Chloride | 3 hours |
| MGHQB8 | Magnaflux | negative |
|  | Nitric Acid (15\%) | $<5 \mathrm{mins}$ |
|  | Wash H 2 O |  |
| MMNK64 | Acid Etch Method | APPROXIMATELY 2 MINUTES |
| MRK49M | Ecid[sic] Etching | +/-30 min |
| MU9FAD | Acidic Ferric Chloride | $<5 \mathrm{sec}$ |
|  | Ferric Chloride (non acidic) |  |
|  | Davis Reagent | $<5 \mathrm{sec}$ |
| MVZCY8 | Chemical Etching |  |
|  | 25\% Sulfuric Acid | 20-30 seconds (Each application) |
|  | Phosphoric / Nitric Acid | 20-30 seconds (Each application) |
|  | 5\% Nitric Acid | 20-30 seconds (Each application) |
|  | Ferric Chloride | 20-30 seconds (Each application) |
| MWYLQ4 | Acid Etch Method | 1 MINUTE |
|  | AQUA REGIA |  |
| MZYPKC | Primarily phosphoric/nitric occasionally acidic/ferric chloride to darken | Applied with a cotton tip applicator leaving the acid on the area for 10-15 seconds then wiping the area with a kimwipe. |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| N6MWFP | Acid Etch Method | Total: 88 minutes |
|  | Ferric Cloride | 38 minutes |
|  | Acidic Ferric Chloride | 50 minutes |
| N7HFGQ | Acidic Ferric Chloride | One minute |
|  | Ferric chloride 19\% | One minute |
|  | Sodium hydroxide 13.5\% | One minute |
| NF84EE | Acidic Ferric Chloride | 5-10 Min |
|  | Acid Etch Method | 5-10 Min |
| NFJZJJ | Fry's Reagent | ten seconds at a time over 45 minutes |
| NMGRQR | Ferric Chloride | approx 15 sec . |
|  | NaOH | approx 10 sec . |
|  | Ferric Chloride | approx 20 sec . |
|  | Nitric Acid 25\% | approx 15 sec . |
|  | NaOH | approx 10 sec . |
|  | Nitric Acid | approx 15-20 sec. |
| NQ64RH | Acidic Ferric Chloride | 10 minutes |
|  | Ferric Chloride | 10 minutes |
|  | Sodium Hydroxide | 10 minutes |
|  | Nitric Acid | 10 minutes |
|  | MagnaFlux | 2 minutes |
| NRW34V | Phosphoric Acid/ 10\% Nitric Acid | 5 minutes |
| NTUDYY | Phosphoric/Nitric Acid | 15 minutes |
|  | Dilute Nitric Acid | 15 minutes |
| NWTRK3 | Ferric Chloride | $\sim 5$ seconds between each swipe |
|  | Nitric Acid 25\% | $\sim 5$ seconds between each swipe |
| NX7GP2 | Phosphoric Nitric Acid and 25\% Nitric Acid | 10 minutes and 30 seconds |
| PAF8BA | Phosphoric/nitric acid | 1 to 5 minutes per application |
|  | 25\% nitric acid | 1 to 5 minutes per application |
|  | Ferric chloride solution | 1 to 5 minutes per application |
|  | DI water rinses |  |
| PGC969 | Ferric Chloride/Acidic Ferric Chloride | 1-2 minutes |
|  | Acidic Ferric Chloride/25\% Nitric Acid | 2-3 minutes |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| PJXXMZ | MagnaFlux | 1 hour |
|  | Acido Nitrico 5\% | 10 minutes |
|  | Fry's Reagent | 10 minutes |
| PL4HDK | Flicks | +/-10 mins |
|  | Nitric Acid | +/-10 mins |
| PP8ZPV | electro-acid etch | +/- 15 minutes |
| PWLY6R | Chemical treatment using $5 \% \mathrm{NaOH}$ solution for 15 minutes follow by | 15 minutes |
|  | Electrochemical treatment using Cupric Ammonium Chloride solution for 8 minutes | 8 minutes |
| PZKB4D | Ferric Chloride | Several brief applications (not timed) |
|  | Acidic Ferric Chloride | Several brief applications (not timed) |
|  | Fry's Reagent | Several brief applications (not timed) |
|  | 25\% Nitric Acid Solution | Several brief applications (not timed) |
| Q2A4B2 | Acidic Ferric Chloride | 5-10 seconds |
|  | Acid Etch Method | 5-10 seconds |
| Q7TYET | MagnaFlux | 10 seconds, one time |
|  | Acidic Ferric Chloride | 10 seconds, two times |
|  | Phosphoric Acid/Nitric Acid | 2-3 seconds, multiple applications |
|  | Nitric Acid | 2-3 seconds, multiple applications |
| Q9XLV4 | HF based solution | 15 minutes |
| QA9NH3 | Acidic Ferric Chloride | $4-5 \mathrm{sec}$ |
|  | Ferric Chloride | $4-5 \mathrm{sec}$ |
|  | Aluminum Solution | $4-5 \mathrm{sec}$ |
|  | Water | 2 sec |
|  | Acidic Ferric Chloride | $4-5 \mathrm{sec}$ |
|  | Aluminum Solution | $4-5 \mathrm{sec}$ |
|  | Water | $4-5 \mathrm{sec}$ |
| QAT6X7 | 25\% Nitric Acid | $<5$ minutes |
|  | 30\% Sulfuric Acid | 1-2 minutes |
|  | Fry's Reagent | < 1 minute |

TABLE 4

| WebCode | Method | Time |
| :--- | :--- | :--- |
| QD9ZD3 | Acidic Ferric Chloride w/Ferric Chl. | $1-3$ mins |
|  | Acidic Ferric w/ 25\% Nitric Acid | $1-3$ mins |
|  | Aluminum Solution (25\% Sulfuric  <br>  Acid) | $1-2$ mins |
|  | Acidic Ferric w/ 25\% Nitric Acid | $1-3$ mins |
|  | Polishing then Acid Ferric w/ 25\% | $1-5$ mins |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| RZWX6Q | Polishing |  |
|  | Acid Etch - Chromic Acid | 1 hour total |
|  | Visual |  |
| T6PDDX | Ferric Chloride | $\sim 5$ seconds |
|  | Diluted FeCl (50/50 mixture) | $\sim 10$ seconds |
|  | 10\% Sodium Hydroxide | $\sim 1$ minute |
| TC8L4F | Acidic Ferric Chloride | 2-6 seconds each time ( $\sim 15$ times total) |
|  | NaOH | $2-3$ seconds at a time ( $\sim 5$ times total) |
|  | 25\% Nitric Acid | $2-3$ seconds at a time ( $\sim 5$ times total) |
|  | Ferric Acid | $2-6$ seconds each time ( $\sim 10$ times total) |
| TKZ48P | Acid Etch |  |
|  | Acid Etch | For about +/-5 min |
| TQ7MH2 | Acidic Ferric Chloride | 15 minutes |
| TXN2G2 | Acidic Ferric Chloride | 7-10 MINUTES |
|  | Fry's Reagent | 3-5 MINUTES |
|  | AQUA-REGIA | 10-15 MINUTES |
| TXQQNJ | Nitric acid | +/-3 min |
| U2YFFQ | Fry's Solution | $<5$ minutes |
| U8BEXC | Modified Fry's Reagent | 1 min . |
|  | Nickels \& Alloys Reagent | 1 min . |
| UDWAYF | Electro-acid etching process was used to recover obliterated serial number | Took about +/-15 minutes |
| UJ34VW | Acidic Ferric Chloride | less than 60 seconds |
|  | 25\% Nitric Acid | less than 60 seconds |
|  | Phosphoric Nitric Acid | less than 60 seconds |
| UP7YQP | Acid Etch | +/-1 minute |
|  | Electro-magnetic method | +/-2 minutes |
| UVWA2N | Acid Etch Method |  |
| UYWKWN | MagnaFlux |  |
|  | 10\% Sodium Hydroxide | 3 min |
|  | Ferric Chloride | 5 min |
|  | Polish |  |
|  | Ferric Chloride | 5 min |
|  | 10 \% Sodium Hydroxide | 2 min |
|  | 50\% Nltric Acid | 5 min |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| UZNR37 | Nitric Acid 10\% | 2 to 3 minutes |
|  | Nitric Acid 25\% | 1 to 2 minutes |
|  | Acidic Ferric Chloride | Less than 1 minute |
| V62RZX | Ferric Chloride | 3 min |
|  | 25\% Nitric Acid | 5 min |
|  | 5\% Nitric Acid | 5 min |
| VE976K | Acid Etch Method | At interval of 1-2 minutes |
| VJZQV8 | Acid Etch Method | less than 1 min . per application |
|  | Ferric Chloride | less than 1 min. per application |
|  | Acidic Ferric Chloride | less than 1 min . per application |
|  | 25\% Nitric Acid | less than 1 min . per application |
|  | Acetone | less than 30 seconds |
|  | Gun oil | not removed |
| VM3CZ8 | MagnaFlux |  |
|  | Acid Etch Method | 10-15 minutes |
| VP6ABZ | Acid Etch Method | 5-minutes |
| VRA33D | Ferric Chloride | $\sim 5$ minutes |
|  | Nitric Acid | $\sim 5$ minutes |
|  | Ferric Chloride | $\sim 1$ minute |
|  | Nitric Acid | $\sim 5$ minutes |
| VYBGNE | Electro-acid etching process | +/- 15 minutes |
| VZ68UT | Acid Etch Method | different acids, all in all about 15 minutes |
| W2F8QM | Acidic Ferric Chloride | 10-15 seconds, |
|  | Phosphoric /Nitric Acid (Step 1) | 20 seconds |
|  | Nitric Acid (Step 2) | 20 seconds |
|  | Acidic Ferric Chloride | 10-15 seconds |
|  | Phosphoric Acid/Nitric Acid (Step 1) | 20 seconds |
|  | Nitric Acid (Step 2) | 20 seconds |
| W4LRF9 | Acid Etch Method | not long apply, swipe, remove |
| W8L3AM | Acidic Ferric Chloride | several applications approx. 3 minutes each |
| WBJJMG | Fry's Reagent | 1 min |
|  | Acidic Ferric Chloride | 1 min |
|  | Acid Etch Method | 15 mins |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| WCARLK | Acidic Ferric Chloride | 10 seconds |
|  | Ferric Chloride | 10 seconds |
|  | Ferric Chloride | 20 seconds |
|  | Ferric Chloride | 30 seconds |
|  | Ferric Chloride | 30 seconds |
| WD8VN9 | Acidic Ferric Chloride | Used over a span for 4 days |
|  | Ferric Chloride | Used over a span for 4 days |
|  | Fry's reagent | Used over a span for 4 days |
| WFA3HK | Ferric Chloride | 30 mins |
|  | Diluted 25\% nitric acid | 10 mins |
| WGAUYZ | Acidic Ferric Chloride | $\sim 25$ min total restoration time |
|  | Nitric Acid |  |
| WL22EN | Acid Etch Method | Approximately three minutes total, in approximately one minute increments |
| WL79U3 | Acidic Ferric Chloride | 10 intervals of 15 seconds each |
| WQWTP7 | Ferric Chloride | Not recorded. Estimated 1 to 3 minutes. |
|  | Acidic Ferric Chloride | Not recorded. Estimated 1 to 3 minutes. |
|  | Nitric Acid | Not recorded. Estimated 30 to 60 seconds. |
| WRRBR8 | Acidic Ferric Chloride | Diluted with Ferric Chloride reagent 5 min. |
|  | Ferric Chloride |  |
| WTK8DF | Electro-acid etching process | +/-15 minutes |
| WTMY7W | Ferric Chloride | 10 seconds |
|  | Acidic Ferric Chloride | 10 seconds |
|  | 25\% Nitric Acid | 10seconds |
| WXJHC2 | 25\% Nitric Acid | Less than 5 minutes |
| XA4J3R | Ferric Chloride | $\sim 10-15$ seconds at a time over $\sim 90$ minutes |
|  | 25\% Nitric Acid | $\sim 10-15$ seconds at a time over $\sim 5$ minutes |
| XNM3Z3 | sodium hydroxide | 10 minutes |
|  | ferric chloride | 20 minutes |

TABLE 4

| WebCode | Method | Time |
| :---: | :---: | :---: |
| XQEYNW | Acidic Ferric Chloride | 1 min |
|  | Ferric Chloride | 1 min |
|  | 10\% Sodium Hydroxide | 1 min |
|  | Aluminum Solution | 30 sec |
|  | Acidic Ferric Chloride | 3 min |
|  | 10\% Sodium Hydroxide | 5 min |
|  | Solution \#2 | 2 min |
| Y8DM7L | Acid Etch | have to remove the acid with cotton wool |
| Y96TC4 | Modified Frys Reagent | 1 min |
|  | Nickels \& Alloys Reagent | 1 min |
| YBWP2X | Acid Etch Method | c. 3 minutes |
|  | Acid Etch Method | c. 30 seconds |
| YKDLTH | Acidic Ferric Chloride | 5 minutes |
|  | Acidic Ferric Chloride | 2 minutes |
|  | Nitric Acid | 1 minute |
|  | 10\% Sodium Hyrdoxide | 2 to 3 minutes |
|  | Nitric Acid | 1 to 2 minutes |
| YKNYU2 | Turner's Reagent | 30 minutes total |
| YMPC8T | MagnaFlux |  |
|  | Acid Etch Method | one minute |
| YXBFER | Acidic Ferric Chloride | less than a minute |
|  | 25\% Nitric Acid | less than a minute |
|  | Phosphoric Nitric Acid | less than a minute |
| Z9JKYT | Ferric Chloride | 30 minutes |
|  | 10\% Sodium Hydroxide | 30 minutes |
|  | 25\% Nitric Acid | 45 minutes |
| ZAXZG2 | Acidic Ferric chloride | $\sim 5$ minutes |
|  | 25\% Nitric Acid | $\sim 10$ minutes |
| ZDHU9V | Solution A | Used multiple times for between 1-30 sec. |
|  | Fry's Reagent | Used multiple times, but only 1-5 sec. at a time |
|  | 10\% Sodium Hydroxide | Used multiple times fro[sic] between 1-30 sec. |
|  | Acidic Ferric Chloride | Used once for 1-5 sec. |
| ZE9ZPL | Acidic Ferric Chloride | 5 times for $5-10$ seconds |

TABLE 4

| WebCode | Method |  | Time |
| :---: | :---: | :---: | :---: |
| ZJMXW9 | Acidic Ferric Chloride |  | 20 swabs |
| ZL47ZE |  |  |  |
| ZX3V4U | Acid Etch Method |  | seconds to several minutes, depending on acid used |
| ZYVNYD | Acid Etch Method |  | Total minutes: |
|  | Ferric Chloride |  | 15 minutes |
|  | Acidic Ferric Chloride |  | 25 minutes |
| ZZWCN4 | Acid Etch Method |  | 30-40 mins |
| Response Summary |  |  | Participants: 302 |
| Recovery Methods |  |  |  |
| Chemical Processing: 276 |  |  |  |
| Magnetic Processing: 15 |  |  |  |
| None: 1 |  |  |  |

## Additional Comments

## TABLE 5

## Additional Comments

26JQKN Initially tried two other acid etching methods which did not produce any numbers or characters. Then was successful with the Phosphoric/Nitric followed by $25 \%$ Nitric acid with some rubbing of the etched surface with a cotton swab moistened with the $25 \%$ Nitric acid solution. Initial attempts were: 1. Phosphoric acid 2 minutes then 5\% Nitric Acid. Obliterated surface was re-polished after this attempt. 2. Ferric Chloride diluted to approximately 50\% allowed to etch for approximately 2 minutes. Neither of the first two methods produced any visible numbers.

2B8DD9 If polishing was done it must be indicated.
2ZQHH3 Note to quality and CTS teams: I disagree that the test sample (to be restored) has the type of material on the envelope. IE: "ZINC BAR STOCK". As experts we should be able to identify the type of material (ferrous or non ferrous), we actually have a method (M7 V1) for telling the difference in materials which we need to do before we commence a Forensic restoration.

32BX2A The surface was sanded until it was mirror-like. The sequence was available to watch with the ferric chloride, the other recovery methods were used for confirm the "J" and the "4".

33R6BX The obliteration was very deep. A shallower obliteration would be appreciated.
3RTGBN Photographs taken prior to beginning, throughout, and after completion of the examination.
4LE34A The piece of aluminum bar stock submitted, was included as a reference for size, shape and positioning of the stamped alphanumeric characters used in the obliterated serial number in the zinc bar stock, described in the ltem 1.

4R2QYE D7 27, D7)274, D7J274. The Sample 1 was photographed and packaged for submission upon completion.

6HCJJM The method "Serial Number Restoration" is accredited according to ISO 17025.
6TVVUY The aluminum bar stock with stamped structure guide and Item 1 (Zinc bar stock with obliterated/restored serial number) were returned to their original envelopes and will be retained in the Firearms Section of the Lab.

88QYAA The visual examination of the item revealed the presence of filing marks on the centre of the bar.

8XYGXM The number in third position could not be completely resolved. It appeared that one side of the numeral was not as deeply impressed in the bar stock as the remaining numerals in the serial number.

96T7RW In casework, there is no standard to use as a guide. There is no need to include the Aluminum Standard.

9E3B6M running water to stop process at each step of analysis sample preparation picturess[sic] were taken

9Q3Y6W High Zinc content made it difficult to view the restored characters due to the Zinc reaction with the reagent $(\mathrm{HCL})$ used. The [Laboratory] cast surfaces prior to destructive testing, ie the use of Reprosil to capture tool marks, partial characters etc, prior to chemical etching. The test would not allow for the addition of that particular methodology from the drop down box.

ACA28Q The serial number was very thin.
AXBZT4 The area with obliterated characters in Item 1 was removed with the use of milling method. The material of Item 1 is paramagnetic.

BWCAZU I would have made a cast of the tool marks had this been an actual case.
BWXQFX Item 1A: zinc bar stock. Item 1B: aluminum standard plate.
CAV438 A digital images was captured after recovery of the obliterated serial No.
DHNT8E After restoration, the area was cleaned with water and swabbed with gun oil.
DRF6JV I used the method above numerous times to try and retrieve the serial number.
E7QG4Z $\quad 10 \% \mathrm{NaOH}$ used for the last 5 minutes of acid etching.
EQU347 After chemical processing the surface was rinsed with water. A small amount of oil was then placed on the surface in order to prevent further reaction.

ERB8L6 Initial polishing with a Dremel and Flitz polishing compound was done lightly and only to remove the apparent milling marks. After a few initial applications of reagents to the serial number area, some vague characters appeared. The location of six characters could be determined, with POSSIBLE characters identified as either "B _ 274 " or "8 _ 27 4". Due to insufficient clarity, the legibility of these characters never reached a threshold that would be considered reportable. The possible "4" remained the most persistent, as the other characters would fade and reappear at various times in the process. Additional polishing was performed only to clean the surface of oxidation in order to minimize the amount of material removed. Less aggressive acid etching techniques ( $25 \%$ nitric acid) and electrolytic techniques employing $10 \%$ sodium hydroxide provided the best treatment results. Griffin's reagent appeared overly aggressive, as did Fry's (which was only briefly applied).

F4JCCX A magnet used to determine if the piece of metal received for analysis it is magnetic or not, to determinate what restoration methodology used.

FR6CHP A cast was made of the toolmarks on Exhibit 1 and marked 1T1.
FTX96X A piece of aluminum bar stock submitted, was used as a reference standard.
FV4TTX The obliterated serial number was restored with phosphoric/nitric acid and $25 \%$ nitric acid.
GBXHLA Cotton swabs seem to be detrimental to restoration effort. Only by allowing the Nitric Acid to pool would characters begin to restore. Characters were difficult to observe when bar stock was dry; final determination of characters restored were done with Nitric Acid puddled on the restored area. Swab almost removed restored characters entirely at this point.

GCAAGC Prior to restoration attempt it was clear that one area was ground down in a very uniform width and depth. With the naked eye, and at the right angle I could see the area that I believed to have had the serial number. Using the Dremel tool with a jeweler's polishing wheel, I polished the area smooth. Soon after the application of $25 \%$ Nitric Acid and Phosphoric Acid (2 step) the serial number was observed.

GF7XVX Recovery method used during examination: acidic method non ferric

## Additional Comments

GJAW7Q Sanding done by hand until obliterated area was smooth to the touch. No numbers were visible after sanding. Fry's was diluted down to $25 \%$. Application of the acids was done slowly with a pause between each application to watch for the appearance of numbers. No numbers appeared.

H24RGT The aluminium bar stock used as the reference to be the same material, zinc bar stock, as the exhibit submitted, if possible. A second reference bar from which metal could be removed to reproduce the action that occur on the exhibit bar stock, in this occurrence. After the acid etching, drops of oil were pored on the area of the retrieved characters to assist with the visible confirmation of the retrieved number under the desk-lamp magnifier.

J46GW8 The item 1 was first given to department of physics in order to verify the CTS information about the material composition. Examinations via scanning-electron-microscope confirmed that item 1 consists of a zinc-aluminum alloy (Content: 96 percent zinc and 4 percent aluminum). Little experience in serial number restoration with stamped characters in almost pure zinc led us to do some experiments. For this purpose we cut off an irrelevant piece of the zinc bar stock and put some stamped characters into its surface. Then the surface was grinded until the characters disappeared. After grinding and polishing the surface again, an acid method was selected by using different reagents for example "NaOH", "Alu1", "Alu2" und "hydrochloric acid". The restoration only succeeded with hydrochloric acid. Afterwards the relevant area of item 1 's surface was treated in the same way (grinding-> polishing-> chemical etching-> record). Using the aluminum bar as a reference for size, shape and positioning of the stamped alphanumeric characters used in the serial number was very helpful. The obliterated serial number appeared weakly as D 7 J 274.

J7FAYL This was a particularly easy restoration. The characters appeared very quickly after application of chemicals. It appears that the characters were stamped deeply, however a relatively small amount of material was milled away.

JEQCKM The zinc metal substrate was an unusual one which this laboratory had not encountered before. The reagents used were not previously utilised by this laboratory, this possibly resulting in the incomplete recovery.

JLLHTU This time Zn was used for bar. No magnetic restoration is allowed (diamagnetic)
JYWTKQ Photographs of the serial number recovery were taken and will be retained in the firearms section of the laboratory.

K3K6MZ I HAD NOT ATTEMPTED TO RESTORE A NUMBER ON ZINC ALLOY PREVIOUSLY TO THIS. THE COPPER CHLORIDE HAD A NEGATIVE EFFECT AFTER APPLICATION AND I DECIDED $25 \%$ NITRIC ACID WOULD BE A MORE RELIABLE ALTERNATIVE.

KCV792 Initially used cupric chloride in nitric acid solution but this was ineffective. Progressed to $25 \%$ nitric acid which successfully restored serial number.

KZLGFQ The obliteration of the zinc bar was too deep. During electro-acid etching process no slight indication of any number or figure was visible.

L4HKM4 We usually use a sanding method to smoothen the surface, but it didn't appear necessary in that case. We used the NaOH and HNO 3 solutions alternatively, until the number was fully recovered.

LGPBMQ If the sample is already polished \& sand it must be mentioned to the participants to avoid

LVJZG3 SEALED PAPER ENVELOPE ORANGE WITH RED TAPE "EVIDENCE" WHITH STIKER WHITE JOINED WITH THE INSCRIPTION 2016 CTS FORENSIC TESTING PROGRAM" TEST N ${ }^{\circ}$ 16-5250: SERIAL NUMBER RESTORATION" "SIMPLE PACK : SNR1" AND INSIDE WHICH ARE TWO (2) ON ORANGE PAPER ONE WITH THE METAL PLATE (TEST N ${ }^{\circ} 16-5250$ ITEM 1 ZINC BAR STOCK) AND ON (ALUMINUM STANDARD TEST N ${ }^{\circ}$ 16-5250) WITH A PLAQUE WITH THE FOLLOWING CHARACTERISTICS PRINTED "ABCDEFHJKN 1234567890"

M8UVBR After the apply the Fry's reagent in the superficie and wash with water, was applied by 1 minute a $\mathrm{NaOH} 13.5 \% \mathrm{~m} / \mathrm{v}$, finally wash with water againt.

MD7UTC For this restoration at first I tried polishing with a Dremel type tool and using MagnaFlux but I did not see anything. Even though the bar was non-ferrous I wanted to try a non-destructive method first. During chemical processing I used Acidic Ferric Chloride twice then polished with the Dremel type tool, and then repeated that cycle two times. After each time I would visually inspect the bar and saw no characters. I then tried $25 \%$ Nitric Acid, this seemed a little strong so I polished and went back to Acidic Ferric Chloride. After two more rounds of Acidic Ferric Chloride and polishing I used Sodium Hydroxide and saw no characters. I polished the zinc metal bar again and then used Acidic Ferric Chloride and allowed it to sit for three hours. After three hours I saw the characters "D 7 J 27 4".

MGHQB8 Methods: Serial Number. Magnetic, thermal, and chemical methods may be used for the restoration of serial numbers. Conclusions regarding restored characters are made by visual examination of the restored surface under a variety of lighting conditions. Information regarding the alpha-numeric structure or the general location of serial numbers is obtained when necessary from reference sources or from firearms in the Laboratory's Reference Firearms Collection. Limitations: Serial Number. With the exception of the magnetic method, serial number restoration is a destructive examination and it is possible that the obtained results may not be reproduced in any subsequent examinations. Restored serial numbers are sometimes only visible during a portion of the reconstruction process, and are not necessarily visible at the conclusion of the process.

MVZCY8 Progress on restoration was not from any one chemical. A combination of the acids restored the number.

MZYPKC Once the restoration was complete, I applied acetone and rig lubricant to the area to preserve. The first character restored is an alpha character "D" followed by a numeric "7" alpha "J" and numerics "2" "7" "4".

N6MWFP The aluminum bar stock described in the item 2, was used for Standard reference, for size, shape and positioning of the stamped alphanumeric characters used in the serial number for the zinc bar stock, described in the item 1 .

NFJZJJ The reagent was applied using a cotton swab and a side to side motion for approximately ten seconds at a time. The acid was wiped off and then reapplied using the same method. This was repeated for a duration of approximately 45 minutes.

NWTRK3 After application of Ferric Chloride, Nitric Acid $25 \%$ was applied and alternated with Ferric Chloride. After the serial number was restored and restoration was complete, deionized water
and then gun oil were applied to the restoration area.
PP8ZPV The number was unable to be restored.
Q9XLV4 The sample was cleaned with water.
RKPVUW Phosphoric/Nitric combo worked first time.
RL43FE Recovered characters faded quickly, and offered poor contrast for photo documentation. Nitric Acid was applied for highlighting efforts, but rendered much worse results.

RNQRLD $\quad 10 \%$ Sodium Hydroxide was suggested for zinc alloy and was applied however it did not have a visible reaction with the polished surface and no hint of the digits were visualised. The surface was washed with water, dried and repolished, then 10\% Nitric Acid (also suggested for zinc) was used with good result to recover the serial number. Out of interest, Fry's Reagent was then applied after the surface was cleaned and repolished. There was a very aggressive reaction on the metal surface and turned it immediately black. This was washed off, the area repolished and $10 \%$ Nitric Acid reapplied which confirmed the serial number, a little clearer than before.

RPG3C7 When conducting the analysis, the last 3 numbers became visible quickly (within 2 minutes) and remained clearly visible, followed by the second number, then the first. Both the first and second numbers were never really as clear or as visible as the last three and I could never really see the middle number during the test, only faint possible outlines.

TC8L4F Since the test did not give an idea of what type of digits (alpha or numeric) were in each space, I did not feel comfortable calling the third placement. I believe it to be a J but since the manufacturer did not notify describe the character designations, I could not be certain. The advantage to actual SNR, is more often than not, you as an analyst understand what the character placement will be, based on manufacturer specs.

TXN2G2 PARTIAL RESTORATION OF SERIAL NUMBER ???274. POSSIBLE SERIAL NUMBERS ARE: 071274, 075274.

UDWAYF My observation: there was no need for one to do polishing and sanding. It was already done. Because its could disturb good results of one's etching. That information could have been stated on the report.

UP7YQP The presence of alpha-numeric figures could be detected but can't be read with the surity required. The finding is negative as it can not be inconclusive.

UZNR37 [From Table 4 - Recovery Methods: Time - "These are approximate times"]
VM3CZ8 The metal has a defect in the area of the third character which obscures the top of the character. Due to this defect, the third character could not be recovered.

VYBGNE The exhibit must not be polished too hard. The results cannot be obtained.
VZ68UT After smoothening the surface softly with sand paper we rubbed the surface with acid saturated cotton swabs. The difficulty to reconstruct the serial number of this Zinc bar stock was comparable with the steel bar stocks the years before.

W2F8QM Cotton swabs used in a rubbing motion while chemicals were applied to item 1 .

## TABLE 5

## WebCode

## Additional Comments

WBJJMG Successful chemical etching used was by alternating between two solutions over a 15 minute time period: Solution 1: 2\% Nitric Acid / 98\% Phosphoric Acid. Solution 2: 5\% Nitric Acid solution.

WTK8DF Polishing and sanding is not necessary for this item, it could influence the results.
WTMY7W Exhibit \# la was not examined.
XQEYNW Serial \# restored using chemical etching process.
YKNYU2 The quality of this test was poorly executed. In the first batch, the number was only partially recoverable after 2 days. The second one received had a very rough surface from the machining and had to be ground down quite a bit before the restoration process could begin.

ZYVNYD The aluminum bar stock, was included as a reference for size, shape and positioning of the stamped alphanumeric characters used in the serial number of the zinc bar stock described in item 1 .

Collaborative Testing Services ~Forensic Testing Program
Test No. 16-5250: Serial Number Restoration
DATA MUST BE RECEIVED BY April 18, 2016 TO BE INCLUDED IN THE REPORT
Participant Code:
Webcode:

## Accreditation Release Statement

CTS submits external proficiency test data directly to ASCLD/LAB and ANAB.
Please select one of the following statements to ensure your data is handled appropriately.
$\square$ This participant's data is intended for submission to ASCLD/LAB and/or ANAB.
(Accreditation Release section on the last page must be completed and submitted.)

$\square$
This participant's data is NOT intended for submission to ASCLD/LAB or ANAB.

Please Note: A piece of aluminum bar stock labeled as "Aluminum Standard" was also included in the sample set and is intended as a reference for size, shape and positioning of the stamped alphanumeric characters used in the serial number.

## Items Submitted (Sample Pack SNR1):

Item 1: A piece of zinc bar stock with suspected obliterated serial number.
1.) Please record the restored characters below.

The serial number on this material consists of 6 characters.

Item 1: $\square$
$\qquad$
$\qquad$
$\qquad$
2.) What would be the wording of the Conclusions in your report?
3.) What methods were used to prepare the sample prior to attempts at recovery? e.g. Sanding, Polishing, Visual, etc. (Please describe in order.)

|  | If sanding was <br> done what grit size <br> was used? |
| :--- | :--- | :--- |

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4.) What recovery methods were used during your examination? e.g. Fry's, Acid Etch, MagnaFlux, etc. (Please list in order of use)

If an acidic method was used how long was the acid left on the material?
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## 5.) Additional Comments

Return Instructions: Data must be received via online

## Participant Code:

data entry, fax (please include a cover sheet), or mail by April 18, 2016 to be included in the report.
Emailed data sheets are not accepted.
QUESTIONS?
TEL: $\quad+1-571-434-1925$ ( $8 \mathrm{am}-4: 30 \mathrm{pm}$ EST)
EMAIL: forensics@cts-interlab.com www.ctsforensics.com

ONLINE DATA ENTRY: www.cts-portal.com
FAX: +1-571-434-1937
MAIL: Collaborative Testing Services, Inc. P.O. Box 650820

Sterling, VA 20165-0820 USA

Collaborative Testing Services ~Forensic Testing Program RELEASE OF DATA TO ACCREDITATION BODIES

The following Accreditation Releases will apply only to:
Participant Code: Webcode:
for Test No. 16-5250: Serial Number Restoration
This release page must be completed and received by April 18, 2016 to have this participant's submitted data included in the reports forwarded to the respective Accreditation Bodies.
ASCLD/LAB RELEASE
If your lab has been accredited by ASCLD/LAB and you are submitting this data as part of their external
proficiency test requirements, have the laboratory's designated individual complete the following.
The information below must be completed in its entirety for the results to be submitted to ASCLD/LAB.
ASCLD/LAB Legacy Certificate No. $\quad$ ASCLD/LAB International Certificate No.
Signature
Laboratory Name
Location (City/State)

ANAB RELEASE
If your laboratory maintains its accreditation through $A N A B$, please complete the following form in its entirety to have your results forwarded.

ANAB Certificate No. $\qquad$
Signature and Title Date $\qquad$
Laboratory Name $\qquad$

Location (City/State) $\qquad$

## Accreditation Release <br> Return Instructions <br> Please submit the completed Accreditation Release at the same time as your full data sheet. See Data Sheet Return Instructions on the previous page.

Questions? Contact us 8 am-4:30 pm EST
Telephone: +1-571-434-1925 email: forensics@cts-interlab.com


[^0]:    This report contains the data received from the participants in this test. Since these participants are located in many countries around the world, and it is their option how the samples are to be used (e.g., training exercise, known or blind proficiency testing, research and development of new techniques, etc.), the results compiled in the Summary Report are not intended to be an overview of the quality of work performed in the profession and cannot be interpreted as such. The Summary Comments are included for the benefit of participants to assist with maintaining or enhancing the quality of their results. These comments are not intended to reflect the general state of the art within the profession.

    Participant results are reported using a randomly assigned "WebCode". This code maintains participant's anonymity, provides linking of the various report sections, and will change with every report.

